# Enantioselective synthesis of $\boldsymbol{\beta}$-hydroxy amines and aziridines using asymmetric transfer hydrogenation of $\boldsymbol{\alpha}$-amino ketones 

Aparecida M. Kawamoto $\dagger$ and Martin Wills*
Department of Chemistry, University of Warwick, Coventry, UK CV4 7AL
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Enantioselective transfer hydrogenation of $\alpha$-amino ketones is an effective method for the asymmetric synthesis of $\beta$-hydroxy amines and aziridines.

## Introduction

Aziridines are valuable synthetic reagents and intermediates. In particular they benefit from high reactivity due to the small, strained, nitrogen-containing rings, thus permitting their rapid conversion into a range of derivatives. The synthesis of enantiomerically pure aziridines is a desirable objective since the physiological properties of both the aziridines themselves and the products formed from them are likely to be dependent on the absolute configuration. ${ }^{1}$
Enantiomerically enriched aziridines may be formed by the enantioselective catalysis of the addition of a nitrene onto one face of a prochiral alkene. ${ }^{2}$ However, the most conceptually simple approach to these targets is probably through the cyclisation of an appropriate enantiomerically pure $\beta$-amino alcohol precursor. ${ }^{1}$ Whilst a number of homochiral $\beta$-amino alcohols are available from natural sources such as amino acids, the range of materials is limited. In this paper we describe a simple and expedient route for the enantioselective synthesis of aziridines via $\alpha$-amido ketone reduction followed by cyclisation.

We have previously described the use of enantioselective transfer hydrogenation ${ }^{3-5}$ for the reduction of ketones containing $\alpha$-amido groups (Scheme 1). ${ }^{5}$ In the example shown we first

$(1 S, 2 R)-\mathbf{1}$

$(R, R)-\mathbf{2}$


Scheme 1 Reagents and conditons: i) $0.5 \mathrm{~mol}^{\%}\left[\mathrm{Ru}(\mathrm{cymene}) \mathrm{Cl}_{2}\right]_{2}$ $1 \mathrm{~mol}^{\%} \% \mathbf{1}$, iPrOH, $2.5 \mathrm{~mol}^{2} \% \mathrm{KOH}$, rt. ii) $0.5 \mathrm{~mol}^{\circ} \%\left[\mathrm{Ru}(\text { cymene }) \mathrm{Cl}_{2}\right]_{2}$ $1 \mathrm{~mol} \% \mathrm{2}, \mathrm{HCO}_{2} \mathrm{H}, \mathrm{Et}_{3} \mathrm{~N}$, rt.
employed a catalyst formed by the combination of $(1 S, 2 R)$-cis1 -aminoindan-2-ol 1 with $\left[\mathrm{Ru} \text { (cymene) } \mathrm{Cl}_{2}\right]_{2}$ and propan-2-olKOH as the hydrogen source, ${ }^{5 a}$ which gave a product of $79 \%$ ee. In subsequent studies we found that the combination of $(R, R)$ -

[^0]TsDPEN 2 with the same ruthenium complex, a system first reported by Noyori, ${ }^{3}$ gave superior results (Scheme 1) when used with the formic acid-triethylamine hydrogen source. ${ }^{5 b}$ Since the enantioselectivies of the reduction were high, we considered that this might be a suitable method for the synthesis of aziridines of high enantiomeric excess.

## Results and discussion

The racemic reduction of $\alpha$-amino ketones is generally routinely carried out through the use of sodium borohydride or other hydride reagents. Enantioselective reduction of such substrates by transfer hydrogenation had not been reported prior to our initial studies, and we feared that it may be inhibited by chelation of the metal of the catalyst by the reduction product. Recent work in our group has revealed that the use of an amide or carbamate derivative to protect the amine prevents this and renders the reaction high yielding and selective. ${ }^{5}$ Following conversion of the hydroxy group to a tosylate and deprotection of the amine it is possible to cyclise the product through an $\mathrm{S}_{\mathrm{N}} 2$ mechanism, to give the aziridine in high ee. ${ }^{5 b}$

During the last two decades, the conversion of 2-amino alcohols to aziridines ${ }^{6,7}$ has been achieved by one-pot cyclisations employing triphenylphosphine-carbon tetrachloride-triethylamine ${ }^{8}$ or triphenylphosphine dibromidetriethylamine. ${ }^{9}$ Although the combination of triphenyl-phosphine-diethyl azodicarboxylate (Mitsunobu reagent) has been used for intramolecular dehydrations, ${ }^{10}$ application of this reagent to the synthesis of aziridines has also been reported. ${ }^{11,12}$ According to Pfister, ${ }^{13}$ aziridines are generally obtained successfully if there is at least one substituent attached to either of the two carbon atoms between oxygen and nitrogen.

For the synthesis of N -benzyl-2-phenylaziridine 3 we began with the preparation of $\alpha$-( $N$-benzyl-amino) ketone from readily available and inexpensive bromoacetophenone and benzylamine, followed by tBoc-protection of the NH group to give $\alpha$-( $N$-benzyl- $N$-tBoc)amino ketone 4 in $79 \%$ yield. Reduction of $\mathbf{4}$ using sodium borohydride gave the racemic alcohol $r a c-5$ in $98 \%$ yield. Enantioselective reduction of $\mathbf{4}$ was successfully achieved using both the ( $1 S, 2 R$ )-cis-aminoindanol-propan-2-ol and ( $1 R, 2 R$ )-TsDPEN-formic acid systems. In both cases the enantiomeric excesses (ee) were determined by chiral HPLC. The aminoindanol system afforded $S-(+)-5$ in $60 \%$ yield and $88 \%$ ee, while the $(1 R, 2 R)$-TsDPEN-formic acid system gave $S$-(+)-5 in $74 \%$ yield and $98 \%$ ee. The absolute configuration was confirmed by conversion to the O-tertbutyldimethylsilyl ether $S-(+)-6$ (see later discussion) by treatment of the precursor alcohol 5 with tert-butylchloro-
dimethylsilane and imidazole ( $49 \%$ yield). Deprotection of the nitrogen atom in $S-(+)-5$ gave the amino alcohol $S-(+)-7$ which was the subject of cyclisation studies. Attempts to achieve this via $O$-tosylation followed by base treatment failed, ${ }^{14}$ however the use of Mitsunobu conditions resulted in formation in low yield $(23 \%)$, but high enantiomeric excess, of the aziridine $R-(-)-\mathbf{3} .{ }^{10}$ The ee of the aziridine $\mathbf{3}$ was determined to be $98 \%$ by ${ }^{1} \mathrm{H}$ NMR analysis using $6 \mathrm{~mol} \% \mathrm{Eu}(\mathrm{hfc})_{3}{ }^{15}$ chiral shift reagent. The racemic aziridine was also synthesised in an identical manner ( $27 \%$ yield for the cyclisation step) (Scheme 2).


Scheme 2 Reagents and conditions: i) $0.5 \mathrm{~mol} \%\left[\mathrm{Ru} \text { (cymene) } \mathrm{Cl}_{2}\right]_{2}$ $1 \mathrm{~mol} \% \mathbf{1}$, iPrOH, $2.5 \mathrm{~mol} \% \mathrm{KOH}$, rt. ii) $0.5 \mathrm{~mol} \%\left[\mathrm{Ru} \text { (cymene) } \mathrm{Cl}_{2}\right]_{2}$ $1 \mathrm{~mol} \%$ 2, $\mathrm{HCO}_{2} \mathrm{H}, \mathrm{Et}_{3} \mathrm{~N}$, rt. iii) TFA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. iv) DEAD, $\mathrm{PPh}_{3}$, THF.

In an attempt to identify an improved route, the synthesis of N -tBoc-2-phenylaziridine was also an objective of our study. $N$-Acylaziridine has been prepared according to essentially three synthetic routes from $N$-acyl- $\alpha$-amino alcohols. The first method involves activation of the hydroxy group by converting it into a tosylate or mesylate and subsequent ring-closure by means of a strong base (LiHMDS or NaH ). ${ }^{16}$ The second method is most frequently used and requires the Mitsunobu conditions. ${ }^{17,18}$ Thirdly $N$-acylaziridines have been prepared using the expensive and hazardous diethylaminosulfur trifluoride (DAST). ${ }^{19}$ For this project we employed the straightforward cyclisation method of $N$-tBoc-amino alcohols described by Wessig et al. ${ }^{14 a}$

The synthetic route began with 2-( $N$-tert-butoxycarbonylamino)acetophenone $\mathbf{8}$ and is summarised in Scheme 3. Our


Scheme 3 Reagents and conditions: i) $0.5 \mathrm{~mol} \%\left[\mathrm{Ru}(\text { cymene }) \mathrm{Cl}_{2}\right]_{2}$ $1 \mathrm{~mol} \%$ 2, $\mathrm{HCO}_{2} \mathrm{H}, \mathrm{Et}_{3} \mathrm{~N}$, rt. ii) TFA, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$. iii) DEAD, $\mathrm{PPh}_{3}$, THF.
studies revealed that the monotosylated diamine system was highly efficient at the reduction of $\mathbf{8}$, furnishing alcohol $S-(+)-\mathbf{9}$ in $86 \%$ yield, $[\alpha]_{\mathrm{D}}^{20}+0.8(c=2$ ethanol $)$, and $99 \%$ ee as measured by chiral HPLC. In contrast the ( $1 S, 2 R$ )-cis-aminoindanolRu (II)-propan-2-ol system totally failed in this application. The reasons for this dramatic difference are not fully clear, however we have previously speculated that chelating reduction products
(as might be obtained from the reduction of $\mathbf{8}$ ) may inhibit, and lead to subsequent decomposition of, the aminoalcohol- Ru (II) complex. ${ }^{3}$ In contrast the monotosylated diamine may well form much stronger complexes with the same metal, and be resistant to such chelation-initiated decomposition. The configuration of $S-(+)-9$ was determined by deprotection of the nitrogen atom with TFA to give the known amino alcohol $S-(+)-\mathbf{1 0}^{\mathbf{2 0}}$ in $27 \%$ yield. Conversion of the same sample of $S-(+)-9$ to $S-(+)-6$ with tert-butylchlorodimethylsilane and imidazole, and subsequently with phenyl bromide and sodium hydride, served to confirm the absolute configuration of the product of reduction of $\mathbf{4}$ as described previously (Scheme 2).

Alcohol $S-(+)-9$ was cyclised to the $N$-tBoc-aziridine $R-(-)-$ 11 in $92 \%$ yield and $99 \%$ ee (determined by ${ }^{1} \mathrm{H}$ NMR analysis using $6 \mathrm{~mol} \% \mathrm{Eu}(\mathrm{hfc})_{3}$ chiral shift reagent) through treatment with tosyl chloride and base, ${ }^{14 a}$ thus delivering an efficient synthesis of aziridines in high yield and enantioselectivity. ${ }^{21}$ Cyclisation of racemic ( $\pm$ )-1-phenyl-2-(tert-butoxycarbonylamino)ethanol was also achieved in $92 \%$ yield through treatment with tosyl chloride and base (in order to supply a racemic standard).

2-Methylaziridine has been prepared using the classic method of Wenker, ${ }^{21,22}$ through cyclisation of the sulfate ester of 1-aminopropan-2-ol. In an attempt to improve the synthetic method of formation of 2-methylaziridine, cyclisation with tosyl chloride and KOH was used. Boc-protection of the NH group of $( \pm)$-1-aminopropan-2-ol gave the $( \pm)$-1-tBoc-aminopropan-2-ol 12 in $95 \%$ yield, which was then used in two different reactions. It was i) cyclised to the $N$-tBoc aziridine 13 through treatment with tosyl chloride and base (57\%), and ii) oxidised with pyridium dichromate to give $( \pm)$-1-tBoc-aminopropan-2-one 14 in $88 \%$ yield. Reduction of 14 using the ruthenium based catalyst, with TsDPEN 2 and formic acid, was achieved in $87 \%$ yield and gave a product of $[a]_{\mathrm{D}}^{20}+27.5(c=1$, $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ ). We have assigned, on the basis of the sense of reduction of $\mathbf{8}, S$ configuration to compound $\mathbf{1 2}$. Alcohol $\mathbf{1 2}$ was then cyclised to $N$-tBoc-2-methylaziridine $R$-(-)-13 in $56 \%$ yield, through treatment with tosyl chloride and base (Scheme 4). The


Scheme 4 Reagents and conditions: i) $0.5 \mathrm{~mol}^{2} \%\left[\mathrm{Ru}(\text { cymene }) \mathrm{Cl}_{2}\right]_{2}$ $1 \mathrm{~mol} \% \mathbf{2}, \mathrm{HCO}_{2} \mathrm{H}, \mathrm{Et}_{3} \mathrm{~N}$, rt. ii) $\mathrm{TsCl}, \mathrm{KOH}, \mathrm{THF}$.
resulting aziridine exhibits $[a]_{\mathrm{D}}^{20}-42.2(c=1$, dichloromethane $)$. Since the alcohol is not UV active, the exact enantiomeric excess of the pure compound could not be measured by chiral HPLC. However Wessig et al. ${ }^{14 a}$ have previously referred to the synthesis of enantiomerically pure $(S)$ - $N$-tBoc-2-methylaziridine, which has $[\alpha]_{\mathrm{D}}^{20}+39.2\left(c=1, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$. Since aziridine 13 has $[a]_{\mathrm{D}}^{20}-42.2\left(c=1, \mathrm{CH}_{2} \mathrm{Cl}_{2}\right)$, we can then conclude that it is of high ee and $R$ configuration as expected since the cyclisation process normally occurs with an inversion of configuration.

In view of the importance of 2-hydroxy-2-propylamine derivatives, such as bisprolol $15,{ }^{23}$ as $\beta$-selective adrenoceptor blocking agents ( $\beta$-blockers) used in the treatment of hypertension, angina pectoris, cardiac arrhythmias, glaucoma, and
prevention of migraine attacks, we chose to examine the synthetic approach to such targets through the enantioselective transfer hydrogenation of the corresponding ketone precursor.


The route to 2-hydroxy-2-propylamine and its corresponding aziridine derivative is outlined is Scheme 5. The substrate was


Scheme 5 Reagents and conditions: i) $0.5 \mathrm{~mol} \%\left[\mathrm{Ru}(c y m e n e) \mathrm{Cl}_{2}\right]_{2}$ $1 \mathrm{~mol} \% \mathbf{1}$, iPrOH, $2.5 \mathrm{~mol}^{2} \% \mathrm{KOH}$, rt. ii) $0.5 \mathrm{~mol}^{2} \%\left[\mathrm{Ru}(\text { cymene }) \mathrm{Cl}_{2}\right]_{2}$ $1 \mathrm{~mol} \% 2, \mathrm{HCO}_{2} \mathrm{H}, \mathrm{Et}_{3} \mathrm{~N}$, rt. iii) $\mathrm{TsCl}, \mathrm{KOH}$, THF.
prepared by tBoc-protection of the NH group of allylamine ( $98 \%$ yield), oxidation with $m$-chloroperoxybenzoic acid ( $88 \%$ ), ring opening with phenol (to give racemic 16, $92 \%$ yield). Compound rac-16 was then i) cyclised to rac- N -tBoc-2(phenoxymethyl)aziridine 17 in $66 \%$ yield, and ii) oxidised with pyridine dichromate to afford $\mathbf{1 8}$ in $31 \%$ yield. Reaction of ketone 18 using the ( $1 R, 2 R$ )-TsDPEN-formic acid system gave the corresponding alcohol $R-(-)-16$ in $84 \%$ ee (measured by chiral HPLC) and $86 \%$ yield. The configuration is based on analogy with the synthesis of $(R)$-propranolol described in the next section. In contrast the ( $1 S, 2 R$ )-cis-aminoindanol-$\mathrm{Ru}(\mathrm{II})$-propan- 2 -ol system totally failed in this application. Conversion of the same sample of $R-(-)-\mathbf{1 6}$ to $R-(+)-\mathbf{1 9}$ with tert-butylchlorodimethylsilane and imidazole, followed by benzyl bromide and sodium hydride, was undertaken to confirm the absolute sense of the product of reduction of $\mathbf{2 0}$ as described in the next section.


Alcohol $R-(-)-\mathbf{1 6}$ was then cyclised to the corresponding aziridine $R-(-)-17$ in $63 \%$ yield, through treatment with tosyl chloride and base. The resulting aziridine has a high $[a]_{\mathrm{D}}^{20}(-68.9$ ( $c=1$, ethanol)), and enantiopurity ( $84 \%$ ee). We believe that this represents a competitive, and highly practical approach for the reduction of this class of ketones, which are normally regarded as 'difficult' substrates due to the lack of steric differentiation between the groups flanking the $\mathrm{C}=\mathrm{O}$ group.

Considering the importance of 2-hydroxy-3-phenoxypropylamine derivatives, some alternative routes to produce the corresponding ketone precursors were examined. $(R)-(+)-1-(N-$
butoxycarbonyl- $N$-benzylamino)-2-oxo-3-phenoxypropane 20 was prepared from the reaction of 2-(chloromethyl)allyl phenyl ether with benzyl amine, tBoc protection and ozonolysis. Enantioselective reduction of ketone $\mathbf{2 0}$ was successfully achieved using both $(1 R, 2 R)$-TsDPEN-formic acid and $(1 S, 2 R)-(+)$-cis1 -aminoindan-2-ol-propan-2-ol systems. The $(1 S, 2 R)$-cisaminoindanol system afforded $(R)-(+)-21$ in $99 \%$ yield with $61 \%$ ee, while the $(1 R, 2 R)$-TsDPEN-formic acid system gave $(R)-(+)-\mathbf{2 1}$ in $99 \%$ yield with $60 \%$ ee. That the absolute configurations of $R-(+)-\mathbf{2 1}$ and $R-(+)-\mathbf{1 6}$ were both identical was confirmed by conversion of $R-(+)$ - 21 to the $O$-tert-butyldimethylsilyl ether $R-(+)-19$ by treatment with tert-butylchlorodimethylsilane and imidazole. Deprotection of the nitrogen atom in $R-(+)-21$ gave the corresponding amino alcohol (not illustrated) for which attempts to cyclise using Mitsunobu conditions failed.

Our assignment of the configuration of compound $R-(+)-21$ was based on analogy with the result from a short synthesis of $R$-(-)-propranolol, a known $\beta$-adrenergic blocking agent ${ }^{24}$ (Scheme 6). Reaction of 1-naphthol with epichlorohydrin

from i) $93 \%$ yield, $64 \%$ ee from ii) $85 \%$ yield, $83 \%$ ee
Scheme 6 Reagents and conditions: i) $0.5 \mathrm{~mol} \%\left[\mathrm{Ru}(c y m e n e) \mathrm{Cl}_{2}\right]_{2}$ $1 \mathrm{~mol} \% \mathbf{1}, \mathrm{iPrOH}, 2.5 \mathrm{~mol} \% \mathrm{KOH}$, rt. ii) $0.5 \mathrm{~mol} \%\left[\mathrm{Ru} \text { (cymene) } \mathrm{Cl}_{2}\right]_{2}$ $1 \mathrm{~mol} \% 2, \mathrm{HCO}_{2} \mathrm{H}, \mathrm{Et}_{3} \mathrm{~N}$, rt. iii) TFA, DCM.
afforded ( $\pm$ )-3-(1-naphthyloxy)-1,2-epoxypropane in $61 \%$ yield, which was followed by a ring opening reaction with isopropylamine to give the racemic amino alcohol, ( $\pm$ )- $N$-isopropyl-$N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine 24 in $93 \%$ yield. tBoc-protection of the amino group of this alcohol gave the ( $\pm$ )- $N$-tBoc- $N$-isopropyl- $N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine 22 in $99 \%$ yield. Oxidation of compound $\mathbf{2 2}$ with pyridinium dichromate afforded the ketone, $N$-(tBoc)-iso-propyl- $N$-[2-oxo-3-(1-naphthyloxy)propyl]amine, 23 in $68.5 \%$ yield. Enantioselective reduction of ketone $\mathbf{2 3}$ was successfully achieved using both ( $1 R, 2 R$ )-TsDPEN-formic acid and $(1 S, 2 R)-(+)$-cis-1-aminoindan-2-ol-propan-2-ol systems. The $(1 S, 2 R)$-cis-aminoindanol system afforded $R$ - $(+)$ - $\mathbf{2 2}$ in $99 \%$ yield with $[a]_{D}^{20}-1.9(c=0.89$, ethanol), while the $(1 R, 2 R)$ -TsDPEN-formic acid system gave $R-(-)-22$ in $98 \%$ yield with $[a]_{D}^{20}-2.92(c=0.96$, ethanol). Deprotection of the amine gave the amino acid, $(R)-(+)-N$-isopropyl- $N$-[2-hydroxy-3-(1naphthyloxy)propyl]amine (propranolol) 24, in $85 \%$ yield and $83.0 \%$ ee as measured by chiral HPLC, with $[a]_{\mathrm{D}}^{20}+5.19$ ( $c=1.6$, ethanol) for $(1 R, 2 R)$-TsDPEN-formic system, and $93 \%$ yield, $64 \% \mathrm{ee}$, with $[a]_{\mathrm{D}}^{20}+6.26(c=0.74$ ethanol) for $(1 S, 2 R)-c i s-$ aminoindanol system.
According to the literature, ${ }^{24} S$-( - )-propranolol has $[a]_{D}^{20}$ -23.2 ( $c=1.08$, ethanol). Therefore, comparing the values of $[a]_{D}^{20}$ of both compounds, we can conclude that our route for the enantioselective reduction of ketone 22 leads to $R-(+)$ propranolol 24. By confirming the configuration of $R-(+)$ -
propranolol, and considering the steric and electronic similarity of the groups OPh with Onaphthyl, and benzyl with isopropyl, we have assigned the configuration of compound 16 and consequently compound 21, as also $R$-configured.

## Conclusion

In conclusion, we have demonstrated that the use of monotosylated diamine-formic acid-triethylamine and cis-1-amino-indan-2-ol-propan-2-ol systems are highly effective at the enantioselective reduction of tBoc-protected $\alpha$-aminoketones and this process provides an efficient method for the enantioselective syntheses of enantiomerically enriched $\beta$-amino alcohol and aziridines.

## Experimental

All reactions, unless otherwise stated, were run under an atmosphere of nitrogen. Room temperature refers to ambient temperature $\left(20-22^{\circ} \mathrm{C}\right), 0{ }^{\circ} \mathrm{C}$ refers to an ice slush bath and $-78^{\circ} \mathrm{C}$ refers to a dry ice-acetone bath. Heated experiments were conducted using thermostatically controlled oil baths. Reactions were monitored by TLC using aluminium backed silica gel 60 (F254) plates, visualised using UV 254 nm and phosphomolybdic acid, ninhydrin or potassium permanganate dips as appropriate. Flash column chromatography was carried out routinely using $60 \AA$ silica gel (Merck). Reagents were used as received from commercial sources unless otherwise stated. Acetophenone was purified by short path distillation before use. Formic acid-triethylamine ( $5: 2$ molar) is a commercially available azeotrope (Fluka). NMR spectra were recorded on a Bruker DPX ( 300 MHz ) spectrometer. The spectra solutions were in deuterated chloroform $\left(\mathrm{CDCl}_{3}\right)$ unless otherwise stated. Chemical shifts are reported in $\delta$ units, parts per million downfield from TMS. Coupling constants $(J)$ are measured in hertz. IR spectra were recorded on a Perkin-Elmer 1310 FTIR instrument using sodium chloride plates. Mass spectra were recorded on a 7070 E VG mass spectrometer. Melting points were recorded on a Stuart Scientific SMP 1 instrument and are uncorrected. Optical rotations were measured with an AA-1000 polarimeter and are given in $10^{-1} \mathrm{deg} \mathrm{cm}^{2} \mathrm{~g}^{-1}$. Determination of enantiomeric excesses by HPLC analysis was achieved using a Waters 501 HPLC pump, Water tuneable absorbance detector, Waters 746 data module and a Daicel Chiral OD $4.6 \times 25 \mathrm{~cm}$ column.

## $\alpha$-( $\boldsymbol{N}$-tert-Butoxycarbonyl- $\boldsymbol{N}$-benzylamino)acetophenone 4

Benzylamine ( $1.02 \mathrm{~g}, 9.5 \mathrm{mmol}$ ) and triethylamine ( $5.06 \mathrm{~g}, 50.0$ $\mathrm{mmol})$ were stirred in dichloromethane $(20 \mathrm{~mL})$ for 30 minutes. 2-Bromoacetophenone (3) ( $1.99 \mathrm{~g}, 10.0 \mathrm{mmol}$ ) in dichloromethane ( 10 mL ) was added dropwise and the reaction stirred for 3 hours. Di-tert-butyl dicarbonate ( $2.18 \mathrm{~g}, 10.0 \mathrm{mmol}$ ) in dichloromethane $(10 \mathrm{~mL})$ was added and the reaction mixture was stirred overnight. Ammonium chloride saturated solution $(30 \mathrm{~mL})$ was added. The phases were separated and the aqueous layer extracted with dichloromethane ( $3 \times 20 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous magnesium sulfate, filtered and the solvent removed in vacuo, to give $\mathbf{4}$ as a yellow oil. Purification by flash chromatography on silica eluting with $5-10 \%$ ethyl acetate-petroleum ether $40: 60$, gave a light yellow solid ( $2.45 \mathrm{~g}, 79.3 \%$ ), mp $66-67^{\circ} \mathrm{C} ; v_{\max }\left(\mathrm{CDCl}_{3}\right) /$ $\mathrm{cm}^{-1} 2960,1703(\mathrm{C}=\mathrm{O}), 1450,1365,1245,1224,1164 ; \delta_{\mathrm{H}}(300$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right)(1: 1$ mixture of NCO rotamers) $1.41(4.5 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{CH}_{3}\right), 1.50\left(4.5 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 4.47(1 \mathrm{H}, \mathrm{s}, \mathrm{CH} H-\mathrm{Ph}), 4.55(1 \mathrm{H}, \mathrm{s}$, $\mathrm{C} H \mathrm{H}-\mathrm{Ph})$, $4.61(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(=\mathrm{O})-\mathrm{CH} H)$, $4.63(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(=\mathrm{O})-$ $\mathrm{C} H \mathrm{H}), 7.22-7.90(10 \mathrm{H}, \mathrm{m}$, Aryl H$)$; $\delta_{\mathrm{c}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) $28.64\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 51.45\left(\mathrm{C}(=\mathrm{O}) C \mathrm{H}_{2}\right), 52.56\left(\mathrm{~N}(\mathrm{Boc}) C \mathrm{H}_{2}\right), 80.72$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 127.82,127.94,128.09,128.31,128.53,129.01$, 129.16, 133.89 (Ar-ipso), 196.56 (C=O); m/z (CI) 326
([M + H] $\left.{ }^{+}, 42 \%\right), 270(46), 227(67), 226$ (100), 169 (22), 120 (56), 106 (40), 91 (34) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 326.1753 . \mathrm{C}_{20} \mathrm{H}_{23} \mathrm{NO}_{3}$ requires $m / z, 326.1756$ ).

## rac-2-( N -tert-Butoxycarbonyl- N -benzylamino)-1-phenylethanol 5

$\alpha$-( $N$-tert-Butoxycarbonyl- $N$-benzylamino)acetophenone 4 $(1.60 \mathrm{~g}, 5.00 \mathrm{mmol})$ was dissolved in 20 ml of $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ ( $9: 1$ ), in an ice bath. Sodium borohydride ( $0.25 \mathrm{~g}, 7.00 \mathrm{mmol}$ ) dissolved in 10 mL of $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}(9: 1)$ was added slowly to the ketone solution. The reaction mixture was allowed to warm up to room temperature. When the reaction was completed (TLC), it was concentrated under reduced pressure. The residue was dissolved in diethyl ether ( 30 mL ) and worked up with ammonium chloride saturated solution $(30 \mathrm{~mL})$. The phases were separated and the aqueous layer extracted with diethyl ether $(2 \times 20 \mathrm{~mL})$. The organic layers were dried over anhydrous magnesium sulfate, filtered and the solvent removed under reduced pressure. The crude compound was purified using medium pressure flash chromatography eluting with $15 \%$ ethyl acetate-petroleum $40: 60$ to give 2 -( $N$-tert-butoxy-carbonyl- $N$-benzylamino)-1-phenylethanol 5 as a white solid ( $1.60 \mathrm{~g}, 98 \%$ ), mp $66-6{ }^{\circ} \mathrm{C}$ (Found: C, 73.24; H, 7.66; N, 4.20. $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{NO}_{3}$ requires C, $\left.73.37 ; \mathrm{H}, 7.70 ; \mathrm{N}, 4.28 \%\right)$; $v_{\text {max }}\left(\mathrm{CDCl}_{3}\right) /$ $\mathrm{cm}^{-1} 3509(\mathrm{OH}), 2927,2365,2357,1669(\mathrm{C}=\mathrm{O}), 1454,1415$, $1246,1164,731,699 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.48\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right)$, $3.32(1 \mathrm{H}$, br d, $J 12.4, \mathrm{CH}(\mathrm{OH}) \mathrm{CH} H), 3.51-3.59(1 \mathrm{H}, \mathrm{m}$, $\mathrm{CH}(\mathrm{OH}) \mathrm{C} H \mathrm{H}), 4.07-4.46(1 \mathrm{H}, \mathrm{m}, \mathrm{CHH}-\mathrm{Ar}), 4.44(2 \mathrm{H}$, br d, $J$ 14.3, $\mathrm{CH} H-\mathrm{Ar}, \mathrm{OH}), 4.88(1 \mathrm{H}, \mathrm{m}, \mathrm{C}(\mathrm{OH}) H), 7.18-7.36(10$ $\mathrm{H}, \mathrm{m}$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.80\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 52.89$ $\left(\mathrm{CH}_{2}\right), 56.04\left(\mathrm{CH}_{2}\right), 74.53(\mathrm{CH}(\mathrm{OH})), 81.42\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 126.21$, 127.36, 127.76, 128.80, 129.01, 129.16, 130.17, 138.34, 142.78 (Ar-ipso), $164.19(\mathrm{C}=\mathrm{O}) ; \mathrm{m} / \mathrm{z}(\mathrm{FAB}) 326\left([\mathrm{M}+\mathrm{H}]^{+}, 14 \%\right), 254$ ( $84 \%$ ), 226 ( $10 \%$ ), 164 ( $22 \%$ ), 154 ( $55 \%$ ), 91 (100).

## 2-(N-Benzylamino)-1-phenylethanol 7

Benzylamine ( $1.29 \mathrm{~g}, 12.1 \mathrm{mmol}$ ) and triethylamine ( $7.6 \mathrm{~g}, 75.0$ mmol ) were stirred in dichloromethane $(50 \mathrm{~mL})$ for 30 minutes. 2-Bromoacetophenone ( $3.0 \mathrm{~g}, 15.1 \mathrm{mmol}$ ) in dichloromethane $(10 \mathrm{~mL})$ was added dropwise and the reaction stirred for 6 hours. Ammonium chloride saturated solution ( 30 mL ) was added. The phases were separated and the aqueous layer extracted with dichloromethane $(3 \times 20 \mathrm{~mL})$. The combined organic layers were dried over anhydrous magnesium sulfate, filtered and the solvent removed in vacuo, to give $\alpha$-( $N$-benzylamino)acetophenone as a yellow oil, which was then dissolved in 20 ml of $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}(9: 1)$, in an ice bath. Sodium borohydride ( $0.86 \mathrm{~g}, 22.6 \mathrm{mmol}$ ) dissolved in 10 ml of $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}$ ( $9: 1$ ) was added slowly to the ketone solution. The reaction mixture was allowed to warm up to room temperature. When the reaction was completed (TLC), it was concentrated under reduced pressure. The residue was dissolved in diethyl ether $(30 \mathrm{~mL})$ and worked up with ammonium chloride saturated solution ( 30 mL ). The phases were separated and the aqueous layer extracted with diethyl ether $(2 \times 20 \mathrm{~mL})$. The organic layers were dried over anhydrous magnesium sulfate, filtered and the solvent removed under reduced pressure. Recrystallisation of crude compound ethyl acetate-petroleum $40: 60$ gave 2-( $N$-benzylamino)-1-phenylethanol as a white solid ( 0.98 g , $35.8 \%), \operatorname{mp} 93-96^{\circ} \mathrm{C} ; v_{\max }\left(\mathrm{CDCl}_{3}\right) / \mathrm{cm}^{-1} 3400,3000,2950,2850$, $2254,1444,903,728 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.40(2 \mathrm{H}$, br s, OH , $\mathrm{N} H), 2.75(1 \mathrm{H}, \mathrm{dd}, J 8.8,12.0, \mathrm{C}(\mathrm{OH}) \mathrm{C} H \mathrm{H}), 2.93(1 \mathrm{H}, \mathrm{dd}$, $J 3.6,12.0, \mathrm{C}(\mathrm{OH}) \mathrm{CH} H), 3.83\left(2 \mathrm{H}, \mathrm{AB}_{\text {system }}, J^{\mathrm{AB}} 13.2\right), 4.73$ (1 H, dd, $J 3.6,8.8, \mathrm{C}(\mathrm{OH}) H), 7.25-7.35(10 \mathrm{H}, \mathrm{m}$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 53.93\left(\mathrm{CH}_{2}\right), 56.94\left(\mathrm{CH}_{2}\right), 72.20(\mathrm{CH})$, 126.22, 127.56, 127.91, 128.51, 128.78, 128.91 (Ar-ipso); m/z (CI) $228\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right), 226$ (45), 212 (20), 210 (66), 150 (10), 120 (65), 108 (36), 91 (55) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 228.1380$. $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{NO}$ requires $m / z, 228.1388$ ).

## rac- $N$-Benzyl-2-phenylaziridine 3

To a solution of ( $\pm$ )-2-benzylamino-1-phenylethanol 7 ( 0.9 g , $3.98 \mathrm{mmol})$ and triphenylphosphine ( $1.36 \mathrm{~g}, 5.20 \mathrm{mmol}$ ) in THF ( 30 mL ), stirred under nitrogen in an ice bath, was slowly added diethyl azodicarboxylate ( $0.9 \mathrm{~g}, 5.20 \mathrm{mmol}$ ) via syringe. The ice bath was removed and the mixture stirred at room temperature for 6 hours. It was then worked up with sodium bicarbonate saturated solution $(20 \mathrm{~mL})$. The phases were separated and the aqueous layer extracted with ethyl acetate $(2 \times 20$ mL ). The organic layers were dried over anhydrous magnesium sulfate, filtered and the solvent removed under reduced pressure. The crude compound was purified using medium pressure flash chromatography eluting with $5 \%$ ethyl acetatepetroleum $40: 60$ to give the aziridine 3 as a pale oil $(0.22 \mathrm{~g}$, $26.5 \%$ ); $v_{\max }($ neat $) / \mathrm{cm}^{-1} 3028,2970,2824,1597,1488,1444$, $1349,1137,1086,1027,823,735 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.84$ $\left(1 \mathrm{H}, \mathrm{d}, J 6.6, \mathrm{CH}_{\mathrm{az}}\right), 1.98\left(1 \mathrm{H}, \mathrm{d}, J 3.2, \mathrm{C} H \mathrm{H}_{\mathrm{az}}\right), 2.49(1 \mathrm{H}$, dd, $J 3.4,6.6, \mathrm{CH}_{\mathrm{az}}$ ), $3.59(1 \mathrm{H}, \mathrm{d}, J 13.8, \mathrm{CH} H \mathrm{Ph}), 3.68(1 \mathrm{H}, \mathrm{d}$, $J 13.8, \mathrm{CH} H \mathrm{Ph}), 7.21-7.38(10 \mathrm{H}, \mathrm{m}, \operatorname{Aryl} \mathrm{H}) ; \delta_{\mathrm{C}}(75.5 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 38.38\left(\mathrm{Ph}-\mathrm{CH}_{2}\right), 41.93\left(\mathrm{CH}_{2 \mathrm{az}}\right)$, $65.17\left(\mathrm{CH}_{\mathrm{az}}\right)$, 126.66, 127.29, 127.38, 128.26, 128.71, 128.77, 139.53, 140.54 (Ar-ipso); $\mathrm{m} / \mathrm{z}(\mathrm{EI}) 210\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right), 208(23), 120(10), 118(20)$, 91 (26) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 210.1309 . \mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}$ requires $m / z$, 210.1304).

## (S)-(+)-2-[ $N$-(tert-Butoxycarbonyl)- $N$-(benzyl)amino]-1-phenylethanol 5

A mixture of ( $p$-cymene) ruthenium(II) chloride dimer ( 0.13 mg , $0.0021 \mathrm{mmol})$ and $(1 R, 2 R)$-TsDPEN $2(0.15 \mathrm{mg}, 0.0042 \mathrm{mmol})$ in a $5: 2$ formic acid-triethylamine mixture $(2.5 \mathrm{~mL})$ was stirred at $28^{\circ} \mathrm{C}$ for 15 min . $2-[N$-(tert-Butoxycarbonyl)- $N$-benzyl-amino]-1-phenylethanone $4(0.27 \mathrm{~g}, 0.83 \mathrm{mmol})$ was added and the solution was stirred at $28^{\circ} \mathrm{C}$ for 24 hours. Then the mixture was filtered through silica and washed with ethyl acetate $(50 \mathrm{~mL})$. The solvent was evaporated under reduced pressure to give the crude compound, which was purified by flash chromatography ( $5 \% \mathrm{v} / \mathrm{v}$ ethyl acetate-petroleum ether $40: 60$ ) to give the product $S-(+)-5$ as a white solid $(0.20 \mathrm{~g}, 74.0 \%)$. The product was determined to be of $98 \%$ ee by HPLC analysis (Chiral OD, hexane-ethanol-diethylamine $=95: 5: 0.1 \quad(0.5$ $\mathrm{mL} \min ^{-1}$ ), $R$ isomer $13.93 \mathrm{~min}, S$ isomer 15.27 min ), mp 64 $65^{\circ} \mathrm{C} ;[a]_{\mathrm{D}}^{20}+3.3\left(c=2\right.$ ethanol); $v_{\max }\left(\mathrm{CDCl}_{3}\right) / \mathrm{cm}^{-1} 3435(\mathrm{OH})$, 3029, 2975, 2931, 1667 (C=O), 1494, 1454, 1414, 1366, 1314, $1246,1165,1123,1086,1060,1028,878,746 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.48\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 3.32(1 \mathrm{H}, \mathrm{d}, J 12.8, \mathrm{CH}(\mathrm{OH}) \mathrm{CH} H)$, $3.52-3.50(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}(\mathrm{OH}) \mathrm{CHH})$, $4.18(1 \mathrm{H}, \mathrm{d}, J 15.6$, $\mathrm{CH} H \mathrm{Ph}), 4.43-4.46(2 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{H}-\mathrm{Ar}, \mathrm{OH}), 4.85-4.96(1 \mathrm{H}, \mathrm{m}$, $\mathrm{C}(\mathrm{OH}) H), 7.18-7.35\left(10 \mathrm{H}, \mathrm{m}\right.$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) $28.81\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 52.89\left(\mathrm{CH}_{2}\right), 56.05\left(\mathrm{CH}_{2}\right), 74.55(\mathrm{CH}(\mathrm{OH}))$, $81.44\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 126.21,127.76,127.94,128.81,129.01,129.16$, 138.32, 142.78 (Ar-ipso); $m / z$ (FAB) 326 ( $[\mathrm{M}+\mathrm{H}]^{+}, 43 \%$ ), 254 (100), 220 (24), 164 (22), 120 (49) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 328.1915$. $\mathrm{C}_{20} \mathrm{H}_{25} \mathrm{NO}_{3}$ requires m/z, 328.1913). The reduction using $(1 S, 2 R)$-1-propan-2-ol was carried out using identical conditions to those reported. ${ }^{4 b}$ The product, $S-(+)-5$ was formed in $60 \%$ yield and $88 \%$ ee as determined by chiral HPLC analysis.

## (S)-(+)-2-( $N$-Benzylamino)-1-phenylethanol 7

To a solution of 2-( $N$-tert-butoxycarbonyl- $N$-benzylamino)-1phenylethanol $S-(-)-5(0.2 \mathrm{~g}, 0.61 \mathrm{mmol})$ in dichloromethane $(2 \mathrm{~mL})$, TFA ( 2 mL ) was added slowly. After 2 hours the solvent was removed and the resulting product was dissolved in ethyl acetate and NaOH solution 0.2 M was added until the $\mathrm{pH}=7$. The product was then extracted with ethyl acetate and the combined organic layers were dried over magnesium sulfate, filtered and the solvent removed in vacuo. Recrystallisation (ethyl acetate-petroleum ether $40: 60$ ) gave ( $S$ )-2-( $N$-benzylamino)-1phenylethanol 7 as a white solid $(0.11 \mathrm{~g}, 78.6 \%$ yield). The
product was determined to be of $99 \%$ ee by HPLC analysis (Chiral OD, hexane-ethanol-diethylamine $=97: 3: 0.1 \quad(0.5$ $\mathrm{mL} \mathrm{min}{ }^{-1}$ ), $S$ isomer $31.97 \mathrm{~min}, R$ isomer 33.19 min ), $\mathrm{mp} 105-$ $107^{\circ} \mathrm{C} ;[a]_{\mathrm{D}}^{20}+33.8\left(c=2\right.$, ethanol); $v_{\max }\left(\mathrm{CDCl}_{3}\right) / \mathrm{cm}^{-1} 3414$, $2248,1637,1452,1201,908,728 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.35$ ( $2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}, \mathrm{NH}$ ), $2.77(1 \mathrm{H}, \mathrm{dd}, J 9.0,12.2, \mathrm{C}(\mathrm{OH}) \mathrm{CHH})$, $2.95(1 \mathrm{H}, \mathrm{dd}, J 3.6,12.2, \mathrm{C}(\mathrm{OH}) \mathrm{CHH})$, $3.86\left(2 \mathrm{H}, \mathrm{AB}_{\text {system }}\right.$, $\left.J^{\mathrm{AB}} 13\right), 4.75(1 \mathrm{H}, \mathrm{dd}, J 3.6,9.0, \mathrm{C}(\mathrm{OH}) \mathrm{H}), 7.20-7.38(10 \mathrm{H}$, m , Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 53.77\left(\mathrm{CH}_{2}\right), 56.69\left(\mathrm{CH}_{2}\right)$, 71.20 (CH), 126.19, 127.75, 127.99, 128.62, 128.81, 128.97 (Ar-ipso); m/z (CI) 228 ([M + H] ${ }^{+}, 100 \%$ ), 226 (7), 210 (10), 120 (36), 108 (11), 91 (11) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 228.1384$. $\mathrm{C}_{15} \mathrm{H}_{17} \mathrm{NO}$ requires $m / z, 228.1388$ ).

## ( $R$ )-(-)-N-Benzyl-2-phenylaziridine 3

To a solution of ( $S$ )-2-benzylamino-1-phenylethanol $5(0.33 \mathrm{~g}$, 1.45 mmol ) and triphenylphosphine ( $0.58 \mathrm{~g}, 2.20 \mathrm{mmol}$ ) in THF ( 15 mL ), stirred under nitrogen in an ice bath, was slowly added diethyl azodicarboxylate ( $0.38 \mathrm{~g}, 2.20 \mathrm{mmol}$ ) via syringe. The ice bath was removed and the mixture was stirred at room temperature for 6 hours. It was then worked up with sodium bicarbonate saturated solution $(10 \mathrm{~mL})$. The phases were separated and the aqueous layer extracted with ethyl acetate $(2 \times 10$ mL ). The organic layers were dried over anhydrous magnesium sulfate, filtered and the solvent removed under reduced pressure. The crude compound was purified using medium pressure flash chromatography eluting with $5 \%$ ethyl acetate-petroleum ether $40: 60$ to give the aziridine $R-(-)-3$ as a pale oil $(0.07 \mathrm{~g}$, $23 \%$ ). The product was determined to be of $98 \%$ ee by chiral shift reagent $\left(\left[\mathrm{Eu}(\mathrm{hfc})_{3}\right], 6 \mathrm{~mol} \%\right)$; $[a]_{\mathrm{D}}^{20}-49(c=2$, ethanol); $v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 3350,3058,3021,2977,1734,1604,1495,1453$, $1259,1172,1028,940,733,697 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.84(1 \mathrm{H}$, d, $\left.J 6.6, \mathrm{CH}_{\mathrm{az}}\right), 1.98\left(1 \mathrm{H}, \mathrm{d}, J 3.2, \mathrm{CHH}_{\mathrm{az}}\right), 2.50(1 \mathrm{H}, \mathrm{dd}$, $\left.J 3.2,6.6, \mathrm{CH}_{\mathrm{az}}\right), 3.69(1 \mathrm{H}, \mathrm{d}, J 13.8, \mathrm{C} H \mathrm{HPh}), 3.58(1 \mathrm{H}, \mathrm{d}$, $J$ 13.8, CHHPh), $7.21-7.42\left(10 \mathrm{H}, \mathrm{m}\right.$, Aryl H); $\delta_{\mathrm{C}}(75.5 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 38.36\left(\mathrm{Ph}-\mathrm{CH}_{2}\right), 41.92\left(\mathrm{CH}_{2 \mathrm{az}}\right), 65.15\left(\mathrm{CH}_{\mathrm{az}}\right), 126.65$, 127.28, 127.38, 128.25, 128.70, 139.51, 140.54 (Ar-ipso); m/z (CI) $210\left([\mathrm{M}+\mathrm{H}]^{+}, 100 \%\right), 209(45), 120$ (7), 108 (17), 91 (24) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 210.1279 . \mathrm{C}_{15} \mathrm{H}_{15} \mathrm{~N}$ requires $\mathrm{m} / \mathrm{z}$, 210.1283).

## ( $\pm$ )-2-( $\boldsymbol{N}$-tert-Butoxycarbonylamino)-1-phenylethanol 9

To a solution of ( $\pm$ )-2-amino-1-phenylethanol $(1.37 \mathrm{~g}, 10.0$ mmol ) in dichloromethane ( 30 mL ), triethylamine ( 1.52 g , 15.0 mmol ) followed by di-tert-butyl dicarbonate ( $2.18 \mathrm{~g}, 10.0$ mmol ) were added. The reaction mixture was stirred overnight then saturated ammonium chloride solution ( 30 mL ) added. The phases were separated and the aqueous layer extracted with dichloromethane $(2 \times 30 \mathrm{~mL})$. The combined organic layers were dried over magnesium sulfate, filtered, and the solvent removed in vacuo to give rac-9 as a white solid ( $2.25 \mathrm{~g}, 95 \%$ ), $\mathrm{mp} 120-121^{\circ} \mathrm{C} ; v_{\max }($ Nujol $) / \mathrm{cm}^{-1} 3362,1675,1531,1276,1166$, $962,957,903,750 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.44\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right)$, 3.18-3.27 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CH} H$ ), $3.39(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 3.42-3.49(1 \mathrm{H}$, $\mathrm{m}, \mathrm{C} H \mathrm{H}), 4.78-4.81(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H), 5.03(1 \mathrm{H}, \mathrm{br} s, \mathrm{~N} H), 7.26-$ $7.35\left(5 \mathrm{H}, \mathrm{m}\right.$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.75\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right)$, $48.73\left(\mathrm{CH}_{2}\right), 74.3(\mathrm{CH}), 80.23\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 126.28,128.16$, 128.87, 142.22 (Ar-ipso), 157.36 ( $\mathrm{C}=\mathrm{O}$ ); $\mathrm{m} / \mathrm{z}$ ( FAB ) 238 ([M + H] ${ }^{+}, 40 \%$ ), 182 (63), 164 (100), 154 (29), 137 (26), 120 (29) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 238.1449. $\mathrm{C}_{13} \mathrm{H}_{19} \mathrm{NO}_{3}$ requires $\mathrm{m} / \mathrm{z}$, 238.1443).

## ( $\pm$ )- N -(tert-Butoxycarbonyl)-2-phenylaziridine 11

To a solution of ( $\pm$ )-2-( $N$-tert-butoxycarbonylamino)-1-phenylethanol $9(0.3 \mathrm{~g}, 1.10 \mathrm{mmol})$ and tosyl chloride $(0.23 \mathrm{~g}, 1.20$ $\mathrm{mmol})$ in dry THF ( 20 mL ) was added $\mathrm{KOH}(0.30 \mathrm{~g}, 5.40$ mmol ) freshly powered (with the help of a ball mill) at room temperature. The reaction mixture was left stirring overnight. It was then dissolved in diethyl ether ( 30 mL ) and filtered. The
solvent was removed under reduced pressure to give pure aziridine $R-(-)-11$ as a pale oil $(0.22 \mathrm{~g}, 92.0 \%) . v_{\max }$ (neat) $/ \mathrm{cm}^{-1}$ 2366, 2336, 2319, 1717, 1655, 1469, 1391, 1317, 1303, 1280, $1253,1229,1153,852,794 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.50(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 2.31\left(1 \mathrm{H}, \mathrm{d}, J 3.58, \mathrm{CH} H_{\mathrm{az}}\right), 2.67(1 \mathrm{H}, \mathrm{d}, J 6.40$, $\left.\mathrm{CHH}_{\mathrm{az}}\right), 3.48\left(1 \mathrm{H}, \mathrm{dd}, J 3.58,6.41, \mathrm{C} H_{\mathrm{az}}\right), 7.31-7.40(5 \mathrm{H}, \mathrm{m}$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.29\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 35.24\left(\mathrm{CH}_{2}\right)$, $39.70(\mathrm{CH}), 81.76\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 126.74,128.15,128.84,137.67$ (Ar-ipso), $162.52(\mathrm{C}=\mathrm{O}) ; m / z(\mathrm{CI}) 220\left([\mathrm{M}+\mathrm{H}]^{+}, 47 \%\right), 181$ (100), 164 (90), 146 (10), 120 (70) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 220.1334$. $\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{NO}_{2}$ requires $m / z$, 220.1338).

## 2-[ $N$-(tert-Butoxycarbonyl)amino]acetophenone 8

( $\pm$ )-2-( $N$-tert-Butoxycarbonylamino)-1-phenylethanol 9 (2.55 $\mathrm{g}, 10.75 \mathrm{mmol})$ was added to an ice cold solution of pyridinium dichromate ( $15.0 \mathrm{~g}, 39.83 \mathrm{mmol}$ ) in anhydrous $N, N$ dimethylformamide ( 50 mL ). The reaction mixture was stirred overnight, then worked up with brine ( 60 mL ). The compound was extracted with diethyl ether $(3 \times 60 \mathrm{~mL})$. The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified by flash chromatography eluting with $10 \%$ ethyl acetate-petroleum ether $40: 60$ to give 2-[ $N$-(tert-butoxycarbonyl)amino]acetophenone $\mathbf{8}$ as a white solid ( $2.23 \mathrm{~g}, 88.1 \%$ ), $\mathrm{mp} 55-58{ }^{\circ} \mathrm{C}$; $v_{\max }\left(\mathrm{CDCl}_{3}\right) / \mathrm{cm}^{-1} 3425$, 2970, 2342, 2247, 1685, 1582, 1488, 1444, 1363, 1217, 1159, $1049,984,903,867 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.48\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right)$, 4.66 ( $2 \mathrm{H}, \mathrm{d}, J 4.52, \mathrm{CH}_{2}$ ), $5.60(1 \mathrm{H}, \mathrm{br}$ s, NH), 7.46-7.97 ( 5 H , m , Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.74\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 47.89$ $\left(\mathrm{CH}_{2}\right), 80.19\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 128.20,129.24,134.30,134.92$ (Aripso), 156.15 (C=O), $193.5(\mathrm{C}=\mathrm{O}) ; \mathrm{m} / z(\mathrm{FAB}) 236$ ( $\mathrm{M}+\mathrm{H}]^{+}$, 23\%), 180 (87), 154 (27), 136 (100), 105 (30), 97 (47) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 236.1284. $\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{NO}_{3}$ requires $m / z$, 236.1287).

## (S)-2-(N-tert-Butoxycarbonylamino)-1-phenylethanol $\boldsymbol{S}$-(+)-9

A mixture of ( $p$-cymene)ruthenium(II) chloride dimer ( 0.34 mg , $0.006 \mathrm{mmol})$ and $(1 R, 2 R)$-TsDPEN $2(0.40 \mathrm{mg}, 0.011 \mathrm{mmol})$ in a $5: 2$ formic acid-triethylamine mixture $(3.0 \mathrm{~mL})$ was stirred at $28^{\circ} \mathrm{C}$ for 15 min . 2 -[ $N$-(tert-Butoxycarbonyl)amino]acetophenone $8(0.50 \mathrm{~g}, 2.13 \mathrm{mmol})$ was added and the solution was stirred at $28^{\circ} \mathrm{C}$ for 24 hours. Then the mixture was filtered through silica and washed with ethyl acetate ( 60 mL ). The solvent was evaporated under reduced pressure to give the crude compound, which was purified by flash chromatography ( $5 \% \mathrm{v} / \mathrm{v}$ ethyl acetate-petroleum ether $40: 60$ ) to give the $S-(+)-$ 9 as a white solid ( $0.44 \mathrm{~g}, 86.3 \%$ ). The product was determined to be of $99 \%$ ee by HPLC analysis (Chiral OD, hexane-ethanol-diethylamine $=95: 5: 0.1\left(0.5 \mathrm{~mL} \mathrm{~min}{ }^{-1}\right), S$ isomer $17.86 \mathrm{~min}, R$ isomer 19.49 min ), $\mathrm{mp} 66-68^{\circ} \mathrm{C}[a]_{\mathrm{D}}^{20}+3.5(c=1$, ethanol); $v_{\max }\left(\mathrm{CDCl}_{3}\right) / \mathrm{cm}^{-1} 3423,2978,2931,1693,1514,1454$, 1392, 1367, 1251, 1170, 1094, 1065, 910, 733, 700; $\delta_{\mathrm{H}}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.43\left(9 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 3.16-3.25(1 \mathrm{H}, \mathrm{m}, \mathrm{CH} H), 3.41-3.47$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CHH}$ ), $3.56(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 4.76-4.79(1 \mathrm{H}, \mathrm{m}$, $\mathrm{C} H(\mathrm{OH})$ ), $5.08(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H), 7.24-7.34(5 \mathrm{H}, \mathrm{m}$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.75\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 48.70\left(\mathrm{CH}_{2}\right), 74.19$ $(\mathrm{CH}), 80.19 \quad\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 126.29,128.14,128.85,142.24$ (Ar-ipso), $157.34(\mathrm{C}=\mathrm{O}) ; m / z(\mathrm{FAB}) 238\left([\mathrm{M}+\mathrm{H}]^{+}, 52 \%\right), 182$ (72), 164 (100), 154 (40), 137 (35), 120 (30) (Found: $[\mathrm{M} \mathrm{+} \mathrm{H}]^{+}$, 238.1448. $\mathrm{C}_{13} \mathrm{H}_{19} \mathrm{NO}_{3}$ requires $m / z, 238.1443$ ). The reduction using ( $1 S, 2 R$ )-1-propan- 2 -ol was carried out using identical conditions to those reported; however no reduction product was isolated. ${ }^{4 b}$

## ( $R$ )-(-)- $N$-(tert-Butoxycarbonyl)-2-phenylaziridine 11

To a solution of ( $S$ )-2-( $N$-tert-butoxycarbonylamino)-1-phenylethanol $S-(+)-9(0.3 \mathrm{~g}, 1.10 \mathrm{mmol})$ and tosyl chloride $(0.23 \mathrm{~g}$, $1.20 \mathrm{mmol})$ in dry THF ( 20 mL ) was added KOH ( $0.30 \mathrm{~g}, 5.40$ mmol ) freshly powered (with the help of a ball mill) at room
temperature. The reaction mixture was left stirring overnight. Then it was dissolved in diethyl ether $(30 \mathrm{~mL})$ and filtered. The solvent was removed under reduced pressure to give a pale oil as a pure aziridine $R-(-)-\mathbf{1 1}(0.22 \mathrm{~g}, 92.0 \%)$. The product was determined to be $99 \%$ ee by chiral shift reagent $\left(\left[\mathrm{Eu}(\mathrm{hfc})_{3}\right]\right.$, $6 \mathrm{~mol} \%),[a]_{\mathrm{D}}^{20}-137\left(c=2\right.$, ethanol); $v_{\max }($ neat $) / \mathrm{cm}^{-1} 3408,3058$, 2977, 2933, 1719, 1663, 1597, 1473, 1318, 1155, 1020, 962, 852, $765 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.44\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 2.26(1 \mathrm{H}, \mathrm{d}$, $\left.J 3.5, \mathrm{CH} H_{\mathrm{az}}\right), 2.62\left(1 \mathrm{H}, \mathrm{d}, J 6.2, \mathrm{CHH}_{\mathrm{az}}\right), 3.42(1 \mathrm{H}, \mathrm{dd}, J 3.5$, $\left.6.2, \mathrm{CH}_{\mathrm{az}}\right), 7.24-7.33\left(5 \mathrm{H}, \mathrm{m}\right.$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) $28.29\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 35.24\left(\mathrm{CH}_{2}\right), 39.70(\mathrm{CH}), 81.82\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right)$, 126.73, 128.15, 128.84, 137.66 (Ar-ipso), 162.54 (C=O); m/z (CI) $220\left([\mathrm{M}+\mathrm{H}]^{+}, 97 \%\right), 181$ (77), 164 (100), 146 (10), 120 (18) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 220.1332. $\mathrm{C}_{13} \mathrm{H}_{17} \mathrm{NO}_{2}$ requires $\mathrm{m} / \mathrm{z}$, 220.1338).

## (S)-(-)-O-(tert-Butyldimethylsilyl)-2-(N-tert-butoxycarbonyl-amino)-1-phenylethanol

(S)-2-(N-tert-Butoxycarbonylamino)-1-phenylethanol $S-(+)-9$ $(0.3 \mathrm{~g}, 1.26 \mathrm{mmol})$ and imidazole $(0.21 \mathrm{~g}, 3.15 \mathrm{mmol})$ were dissolved in 10 mL of DMF and treated with tert-butylchlorodimethylsilane (TBDMSCl) $(0.23 \mathrm{~g}, 1.52 \mathrm{mmol})$. The mixture was stirred at $35-45^{\circ} \mathrm{C}$ until the starting materials were completely consumed (followed by TLC). The reaction was then treated with sodium bicarbonate saturated solution $(20 \mathrm{~mL})$ and extracted with diethyl ether $(3 \times 20 \mathrm{~mL})$. The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified by flash chromatography eluting with $5 \%$ ethyl acetate-petroleum ether $40: 60$ to give the product as a pale liquid $(0.37 \mathrm{~g}, 84.1 \%)$. This was determined to be of $92 \%$ ee by HPLC analysis (Chiral OD hexane-ethanol-diethylamine $=97: 3: 0.1 \quad(0.5 \mathrm{~mL} \mathrm{~min}-1)$, $S$ isomer $7.59 \mathrm{~min}, R$ isomer 9.96 min$),[a]_{\mathrm{D}}^{20}+55.7(c=2$, ethanol); $v_{\max }($ neat $) / \mathrm{cm}^{-1} 3460,3373,2956,2885,1716,1504$, $1435,1390,1252,1171,1098,976,870,700 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 0.00\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)\right)$, $0.14\left(3 \mathrm{H}, \mathrm{s}, \mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)\right)$, $1.01\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.54\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C} H_{3}\right), 3.11-3.20(1 \mathrm{H}$, $\mathrm{m}, \mathrm{CH} H), 3.42-3.56(1 \mathrm{H}, \mathrm{m}, \mathrm{CHH})$, 4.86-4.90 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CH}$ ), 7.35-7.43 ( $5 \mathrm{H}, \mathrm{m}$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-4.65$ $\left(\mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)\right),-4.34\left(\mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)\right), 18.61\left(\mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 26.22$ $\left(\mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 28.81\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 49.50\left(\mathrm{CH}_{2}\right), 74.20(\mathrm{CH}), 79.60$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 126.37,127.82,128.60,142.80$ (Ar-ipso), 156.25 (C=O); $m / z(\mathrm{CI}) 352$ ([M + H] $\left.{ }^{+}, 96 \%\right), 296$ (100), 278 (16), 252 (77), 221 (82), 181 (72).

## ( S )- O -(tert-Butyldimethylsilyl)-2-[ N -(tert-butoxycarbonyl)- N -benzylamino]-1-phenylethanol $S$-(+)-6

To a suspension of sodium hydride ( $0.05 \mathrm{~g}, 1.14 \mathrm{mmol})$ in dry THF ( 10 mL ), ( S )-O-(tert-butyldimethylsilyl)-2-[ N -)tert-butoxycarbonyl)aminoj-1-phenylethanol ( $0.2 \mathrm{~g}, 0.57 \mathrm{mmol}$ ) was added at room temperature and it was stirred for 1 hour. Then a solution of benzyl bromide ( $0.15 \mathrm{~g}, 0.85 \mathrm{mmol}$ ) in dry THF ( 5 mL ) was added slowly. The reaction was left stirring until the starting materials had been completely consumed (followed by TLC). The reaction was then treated with sodium bicarbonate saturated solution ( 20 mL ) and extracted with ethyl acetate $(3 \times 20 \mathrm{~mL})$. The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified by flash chromatography eluting with $5 \%$ ethyl acetatepetroleum ether $40: 60$ to give the protected compound as a pale liquid $(0.25 \mathrm{~g}, 100 \%)$. The product was determined to be of $92 \%$ ee by HPLC analysis (Chiral OD, hexane-ethanoldiethylamine $=97: 3: 0.1\left(0.5 \mathrm{~mL} \mathrm{~min}{ }^{-1}\right), S$ isomer $8.63 \mathrm{~min}, R$ isomer 9.45 min$),[a]_{\mathrm{D}}^{20}+52.3\left(c=2\right.$, ethanol); $v_{\max }($ neat $) / \mathrm{cm}^{-1}$ 3058, 3028, 2955, 2919, 2846, 1700, 1693, 1458, 1440, 1363, 1247, 1167, 1020, 947, 838, 779; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)(1: 1$ mixture of NCO rotamers) $0.00\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)\right), 0.004$
$\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)\right), 0.17\left(1.5 \mathrm{H}, \mathrm{s}, \mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)\right), 0.19(1.5 \mathrm{H}$, $\mathrm{s}, \mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)$ ), $1.02\left(4.5 \mathrm{H}, \mathrm{s}, \mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.06(4.5 \mathrm{H}$, s, $\left.\mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.56\left(4.5 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.65\left(4.5 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, $3.13(0.5 \mathrm{H}, \mathrm{dd}, J 9.1,14.2, \mathrm{CH}), 3.23(0.5 \mathrm{H}, \mathrm{dd}, J 7.9,14.3$, CH), $3.40(0.5 \mathrm{H}$, dd, $J 4.7,14.5, \mathrm{CH}), 3.60(0.5 \mathrm{H}$, dd, J 3.4, $13.8, \mathrm{CH}), 4.46(0.5 \mathrm{H}, \mathrm{d}, J 5.8, \mathrm{H}), 4.48(0.5 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{CH})$, 4.58 ( $0.5 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{CH}$ ), $4.72(0.5 \mathrm{H}, \mathrm{d}, J 5.8, \mathrm{CH}), 5.08$ $(0.5 \mathrm{H}, \mathrm{dd}, J 5.0,8.0, \mathrm{CH}), 5.24(0.5 \mathrm{H}, \mathrm{dd}, J 3.6,9.0, \mathrm{CH})$, 7.25-7.49 ( $10 \mathrm{H}, \mathrm{m}$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 26.31$ $\left(\mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)\right), 28.85\left(\mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 28.96\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 51.81$, $53.25\left(\mathrm{CH}_{2}\right)$, $55.54,56.45\left(\mathrm{CH}_{2}\right), 68.37\left(\mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 73.84$, $74.52(\mathrm{CH}), 80.12\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 126.35,127.03,127.35,127.44$, 127.63, 127.99, 128.47, 128.61, 128.78, 128.82, 138.78, 139.12, 143.36 (Ar-ipso), 156.10 (C=O); m/z (FAB) 442 ( $[\mathrm{M}+\mathrm{H}]^{+}$, $29 \%$ ), 342 (44), 328 (45), 254 (88), 221 (74), 164 (29), 120 (59), 91 (100).

A sample of $S-(+)-6$ with identical data to that cited above was also prepared by the reaction of $S-(+)-5$ with TBDMSCl and imidazole in DMF following the same procedure as that previously described for $S-(+)-9$. The product was formed in $49 \%$ yield and exhibited $[a]_{\mathrm{D}}^{20}+36.0(c=2$, ethanol $)$.

## ( $\pm$ )-1-( N-tert-Butoxycarbonylamino)propan-2-ol 12

To a solution of ( $\pm$ )-1-aminopropan-2-ol $(2.0 \mathrm{~g}, 26.62 \mathrm{mmol})$ in dichloromethane ( 50 mL ), triethylamine ( $3.95 \mathrm{~g}, 15.0 \mathrm{mmol}$ ) followed by di-tert-butyl dicarbonate ( $2.18 \mathrm{~g}, 39.0 \mathrm{mmol}$ ) were added. The reaction mixture was stirred overnight, then saturated ammonium chloride solution ( 30 mL ) was added. The phases were separated and the aqueous layer extracted with dichloromethane $(2 \times 30 \mathrm{~mL})$. The combined organic layers were dried over magnesium sulfate, filtered, and the solvent removed in vacuo. Purification by flash chromatography on silica with $20 \%$ ethyl acetate-petroleum ether $40: 60$, gave the compound rac-12 as colourless liquid of high viscosity ( 4.44 g , $95.3 \%$ ); $v_{\max }($ neat $) / \mathrm{cm}^{-1} 2977,2357,1685,1517,1363,1276$, $1159,1108,976,750 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.17(3 \mathrm{H}, \mathrm{d}, J 6.41$, $\left.\mathrm{CH}_{3}\right), 1.45\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.95-3.04(2 \mathrm{H}, \mathrm{m}, \mathrm{CHH}, \mathrm{OH})$, 3.23-3.29 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{C} H$ ), 3.88-3.90 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{H}$ ), $5.12(1 \mathrm{H}, \mathrm{br}$ $\mathrm{s}, \mathrm{N} H) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 20.32\left(\mathrm{CH}_{3}\right), 28.47\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right)$, $47.91\left(\mathrm{CH}_{2}\right), 67.06(\mathrm{CH}), 79.83\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 164.25(\mathrm{C}=\mathrm{O})$; $m / z(\mathrm{FAB}) 176$ ( $[\mathrm{M}+\mathrm{H}]^{+}, 74 \%$ ), 154 (6), 137 (9), 120 (100), 103 (13) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 176.1286 . \mathrm{C}_{8} \mathrm{H}_{17} \mathrm{NO}_{3}$ requires $\mathrm{m} / \mathrm{z}$, 176.1287).

## ( $\pm$ )- N -(tert-Butoxycarbonyl)-2-methylaziridine 13

To a solution of ( $\pm$ )-1-( $N$-tert-butoxycarbonylamino)propan-2-ol $\mathbf{1 2}(2.0 \mathrm{~g}, 11.41 \mathrm{mmol})$ and tosyl chloride $(3.05 \mathrm{~g}, 15.98$ $\mathrm{mmol})$ in dry THF ( 40 mL ) was added KOH ( 3.10 g , 56.11 mmol ) freshly powered (with the help of a ball mill) at room temperature. The reaction mixture was left stirring overnight. Then it was dissolved in diethyl ether ( 50 mL ) and filtered. The solvent was removed under reduced pressure and the crude compound was purified using flash chromatography eluting with $20 \%$ ethyl acetate-petroleum ether $40: 60$ to give the aziridine rac-13 as a colourless liquid ( $1.27 \mathrm{~g}, 56.7 \%$ ); $v_{\max }$ (neat) $/ \mathrm{cm}^{-1} 3006,2989,2357,1714,1451,1275,1260,1159$, $749 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.22\left(3 \mathrm{H}, \mathrm{d}, J 5.6, \mathrm{CH}_{3}\right), 1.41(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 1.82\left(1 \mathrm{H}, \mathrm{d}, J 3.8, \mathrm{CH} H_{\mathrm{az}}\right), 2.19(1 \mathrm{H}, \mathrm{d}, J 5.8$, $\left.\mathrm{C} H \mathrm{H}_{\mathrm{az}}\right), 2.14-2.43\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H_{\mathrm{az}}\right) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 17.28$ $\left(\mathrm{CH}_{3}\right), 28.12\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 32.41\left(\mathrm{CH}_{2}\right), 33.69(\mathrm{CH}), 80.92$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 162.58(\mathrm{C}=\mathrm{O}) ; m / z(\mathrm{FAB}) 158\left([\mathrm{M}+\mathrm{H}]^{+}, 16 \%\right), 156$ (10), 128 (6), 119 (15), 108 (37), 58 (100) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 158.1181. $\mathrm{C}_{8} \mathrm{H}_{15} \mathrm{NO}_{2}$ requires $m / z, 158.1181$ ).

## ( $\pm$ )-1-( $N$-tert-Butoxycarbonylamino)propan-2-one 14

$( \pm)$-1-( $N$-tert-Butoxycarbonylamino)propan-2-ol $(1.0 \mathrm{~g}, 5.71$ mmol ) was added to an ice cold solution of pyridinium dichromate ( $6.44 \mathrm{~g}, 17.13 \mathrm{mmol}$ ) in anhydrous $N, N$-dimethyl-
formamide ( 20 mL ). The reaction mixture was stirred overnight, then worked up with brine $(30 \mathrm{~mL})$. The compound was extracted with diethyl ether $(3 \times 30 \mathrm{~mL})$. The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified by flash chromatography eluting with $40 \%$ ethyl acetate-petroleum ether $40: 60$ to give $( \pm)-1-(N$-tert-butoxycarbonylamino) propan-2-one 14 as a colourless liquid ( 0.88 g , $88.0 \%) ; v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 3357,2977,2926,1699,1510,1356$, 1283, 1247, 1159, 1071, 955, 882, 779; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right.$ ) $1.45\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.18\left(3 \mathrm{H}, \mathrm{s}, \mathrm{CH}_{3}\right), 4.03(2 \mathrm{H}, \mathrm{d}, J 4.71$, $\left.\mathrm{CH}_{2}\right), 5.24(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 27.49\left(\mathrm{CH}_{3}\right)$, $28.69\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 51.31\left(\mathrm{CH}_{2}\right), 80.23\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right)$, $164.05(\mathrm{C}=\mathrm{O})$; $\mathrm{m} / \mathrm{z}$ (CI) 174 ([M + H] ${ }^{+}, 100 \%$ ), 163 (12), 147 (25) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 174.1127. $\mathrm{C}_{8} \mathrm{H}_{15} \mathrm{NO}_{3}$ requires $\left.m / z, 174.1130\right)$.

## $S-(+)-1-(N-$ tert-Butoxycarbonylamino)propan-2-ol 12

A mixture of ( $p$-cymene)ruthenium(II) chloride dimer ( 0.62 mg , $0.010 \mathrm{mmol})$ and $(1 R, 2 R)$-TsDPEN $2(0.74 \mathrm{mg}, 0.02 \mathrm{mmol})$ in a $5: 2$ formic acid-triethylamine mixture $(3.0 \mathrm{~mL})$ was stirred at $28^{\circ} \mathrm{C}$ for 15 min . ( $\pm$ )-1-( $N$-tert-Butoxycarbonylamino)-propan-2-one $14(0.70 \mathrm{~g}, 4.04 \mathrm{mmol})$ was added and the solution was stirred at $28^{\circ} \mathrm{C}$ for 24 hours. Then the mixture was filtered through silica and washed with ethyl acetate $(60 \mathrm{~mL})$. The solvent was evaporated under reduced pressure to give the crude compound, which was purified by flash chromatography ( $30 \%$ $\mathrm{v} / \mathrm{v}$ ethyl acetate-petroleum ether $40: 60$ ) to give $S-(+)-12$ as a colourless liquid ( $0.62 \mathrm{~g}, 87.3 \%$ ). $[a]_{\mathrm{D}}^{20}+27.5(c=2$, dichloromethane); $v_{\max }$ (neat) $/ \mathrm{cm}^{-1} 3355,2974,2934,1691,1528,1458$, 1393, 1368, 1276, 1252, 1173, 1039, 991, 942, 901, 865, 838; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.18\left(3 \mathrm{H}, \mathrm{d}, J 6.41, \mathrm{CH}_{3}\right), 1.45(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)\right), 2.96-3.05(2 \mathrm{H}, \mathrm{m}, \mathrm{CHH}, \mathrm{OH}), 3.23-3.30(1 \mathrm{H}, \mathrm{m}$, $\mathrm{C} H \mathrm{H}), 3.85-3.95(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{H}), 5.01(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{N} H) ; \delta_{\mathrm{c}}(75.5$ $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) 21.03\left(\mathrm{CH}_{3}\right), 28.75\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 48.35\left(\mathrm{CH}_{2}\right)$, $68.09(\mathrm{CH}), 80.07\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 155.60(\mathrm{C}=\mathrm{O}) ; m / z(\mathrm{FAB}) 176$ $\left([\mathrm{M}+\mathrm{H}]^{+}, 27 \%\right), 155(100), 154$ (100), 136 (80), 120 (50), 107 (31) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 176.1279. $\mathrm{C}_{8} \mathrm{H}_{17} \mathrm{NO}_{3}$ requires $\mathrm{m} / \mathrm{z}$, 176.1287).

## $R-(-)$ - $\mathbf{N - ( t e r t - B u t o x y c a r b o n y l ) - 2 - m e t h y l a z i r i d i n e ~} 13$

To a solution of (+)-1-(N-tert-butoxycarbonylamino)propan-2-ol $S$-(+)-12 ( $1.0 \mathrm{~g}, 5.71 \mathrm{mmol}$ ) and tosyl chloride ( $1.63 \mathrm{~g}, 8.56$ mmol ) in dry THF ( 20 mL ) was added $\mathrm{KOH}(1.60 \mathrm{~g}, 28.55$ mmol ) freshly powered (with the help of a ball mill) at room temperature. The reaction mixture was left stirring overnight. Then it was dissolved in diethyl ether ( 40 mL ) and filtered. The solvent was removed under reduced pressure and the crude compound was purified using flash chromatography eluting with $20 \%$ ethyl acetate-petroleum ether $40: 60$ to give the $R-(-)-13$ as a colourless liquid $(0.50 \mathrm{~g}, 55.7 \%)[a]_{\mathrm{D}}^{20}-42.2$ ( $c=1$, dichloromethane); $v_{\max }($ neat $) / \mathrm{cm}^{-1} 3064,2977,2932$, $1718,1600,1475,1458,1407,1393,1368,1310,1257,1224$, 1154, 1088, 1063, 1028, 940, 908, 867, 831, 769; $\delta_{\mathrm{H}}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.27\left(3 \mathrm{H}, \mathrm{d}, J 5.5, \mathrm{CH}_{3}\right), 1.46\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 1.89$ $\left(1 \mathrm{H}, \mathrm{d}, J 3.8, \mathrm{CH} H_{\mathrm{az}}\right), 2.24\left(1 \mathrm{H}, \mathrm{d}, J 5.8, \mathrm{CHH}_{\mathrm{az}}\right), 2.24-2.49$ $\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{az}}\right) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 17.77\left(\mathrm{CH}_{3}\right), 28.31$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 32.86\left(\mathrm{CH}_{2}\right), 33.95(\mathrm{CH}), 81.29\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 162.58$ (C=O); $m / z$ (FAB) 158 ([M + H] ${ }^{+}, 16 \%$ ), 156 (10), 128 (6), 119 (15), 108 (37), 58 (100) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 158.1181. $\mathrm{C}_{8} \mathrm{H}_{15} \mathrm{NO}_{2}$ requires $m / z, 158.1181$ ).

## $N$-(tert-Butoxycarbonyl)allylamine

To a solution of allylamine ( $3.0 \mathrm{~g}, 52.54 \mathrm{mmol}$ ) in dichloromethane ( 80 mL ), triethylamine ( $6.91 \mathrm{~g}, 68.30 \mathrm{mmol}$ ) followed by di-tert-butyl dicarbonate ( $11.57 \mathrm{~g}, 53.0 \mathrm{mmol}$ ) were added. The reaction mixture was stirred overnight, then saturated ammonium chloride solution ( 50 mL ) added. The phases were separated and the aqueous layer extracted with dichloro-
methane $(2 \times 50 \mathrm{~mL})$. The combined organic layers were dried over magnesium sulfate, filtered, and the solvent removed in vacuo. Purification by flash chromatography on silica with $10 \%$ ethyl acetate-petroleum ether $40: 60$, gave the product as white solid ( $8.1 \mathrm{~g}, 98.1 \%$ ); $v_{\max }\left(\mathrm{CDCl}_{3}\right) / \mathrm{cm}^{-1} 3352,2976,1693$, 1519, 1366, 1250, 1173, 990, 917, 862, 780; $\delta_{\mathrm{H}}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.45\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 3.73-3.75\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}=\mathrm{CH}-\right.$ $\left.\mathrm{CH}_{2}\right), 4.71(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH}), 5.10\left(1 \mathrm{H}, \mathrm{dq}, J 10.2,1.5, \mathrm{CH} H^{c i s}\right)$, 5.18 ( $\left.1 \mathrm{H}, \mathrm{dq}, J 17.1,1.5, \mathrm{CHH}^{\text {rans }}\right)$, $5.84(1 \mathrm{H}$, ddt, $J 10.2,17.1$, $\left.5.5,=\mathrm{CH}-\mathrm{CH}_{2}\right) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.75\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 43.42$ $\left(\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{2}\right), 79.69 \quad\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 116.01\left(\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{2}\right)$, $135.30\left(\mathrm{CH}_{2}=\mathrm{CH}-\mathrm{CH}_{2}\right), 156.17(\mathrm{C}=\mathrm{O})$.

## rac- N -(tert-Butoxycarbonyl)-2,3-epoxypropylamine

A solution of $N$-(tert-butoxycarbonyl)allylamine ( $3 \mathrm{~g}, 19.08$ mmol ) in anhydrous dichloromethane ( 60 mL ) was added dropwise to a stirred solution of $m$-chloroperoxybenzoic acid $(9.87 \mathrm{~g}, 57.24 \mathrm{mmol})$ at room temperature. Then it was refluxed for 6 hours and worked up with sodium bicarbonate saturated solution ( 40 mL ). The compound was extracted with ethyl acetate ( $3 \times 40 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified by flash chromatography eluting with $10 \%$ ethyl acetate-petroleum ether $40: 60$ to give $N$-(tert-butoxycarb-onyl)-2,3-epoxypropylamine as a pale liquid ( $2.75 \mathrm{~g}, 88.3 \%$ ) of high viscosity which crystallised on the bench after one week; $v_{\max }\left(\mathrm{CDCl}_{3}\right) / \mathrm{cm}^{-1} 3451,2980$, 2931, 2252, 1704, 1510, 1454 , 1391, 1367, 1332, 1251, 1170, 1097, 1041, 1020, 908, 849, 735, $648 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.45\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 2.60(1 \mathrm{H}$, dd, $J$ 2.6, 4.5, OC $\left.H \mathrm{HCH}-\mathrm{CH}_{2}\right), 2.78(1 \mathrm{H}, \mathrm{dd}, J 4.5,4.3$, OCHHCH-CH2), $3.07-3.12\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}-\mathrm{CH}_{2}-\mathrm{NH}\right), 3.17-3.29$ $(1 \mathrm{H}, \mathrm{m}, \mathrm{CH} H-\mathrm{NH}), 3.51-3.56(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{H}-\mathrm{NH}), 4.76(1 \mathrm{H}$, $\mathrm{br} \mathrm{s}, \mathrm{NH}) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.73\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 42.01$ $\left(\mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}-\mathrm{CH}_{2}\right), 45.41\left(\mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}-\mathrm{CH}_{2}\right), 51.23\left(\mathrm{O}-\mathrm{CH}_{2}-\right.$ $\left.\mathrm{CH}-\mathrm{CH}_{2}\right), 80.03\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 164.29(\mathrm{C}=\mathrm{O}) ; \mathrm{m} / \mathrm{z}$ (CI) 174 $\left([\mathrm{M}+\mathrm{H}]^{+}, 62 \%\right), 135(100), 118(84), 74(35), 57(7)$ (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 174.1131 . \mathrm{C}_{8} \mathrm{H}_{15} \mathrm{NO}_{3}$ requires $m / z, 174.1130$ )

## rac-1-(N-tert-Butoxycarbonylamino)-2-hydroxy-3-phenoxypropane 16

Potassium hydroxide ( $0.65 \mathrm{~g}, 11.6 \mathrm{mmol}$ ), in water ( 15 mL ), was cooled at $0^{\circ} \mathrm{C}$, then $N$-(tert-butoxycarbonyl)-2,3-epoxypropylamine ( $1 \mathrm{~g}, 5.77 \mathrm{mmol}$ ) and phenol ( $1.08 \mathrm{~g}, 11.6 \mathrm{mmol}$ ) were added to the solution. The reaction was warmed up to room temperature, left stirring overnight, then neutralised with 2 M hydrochloric acid. The compound was extracted with ethyl acetate ( $3 \times 30 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified by flash chromatography eluting with $20 \%$ ethyl acetatepetroleum ether 40 : 60 to give 1-( $N$-tert-butoxycarbonylamino) 2-hydroxy-3-phenoxypropane rac-16 as a pale liquid $(2.49 \mathrm{~g}$, $92.2 \%) ; v_{\max }($ neat $) / \mathrm{cm}^{-1} 3359,3060,2977,2931,1690,1599$, 1587, 1497, 1456, 1391, 1366, 1245, 1170, 1123, 1070, 1044, 993 , $925,855,813,781,754,737,691 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.46$ ( $\left.9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 3.26-3.35\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right), 3.44-3.52(1 \mathrm{H}, \mathrm{m}$, $\mathrm{C}(\mathrm{OH}) \mathrm{H}), 3.90-4.01\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right), 4.08(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}), 5.01$ ( $1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH}$ ), 6.86-7.32 $\left(5 \mathrm{H}, \mathrm{m}\right.$, Aryl-H); $\delta_{\mathrm{C}}(75.5 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 28.74\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right)$, $48.89\left(\mathrm{CH}_{2}\right), 69.66\left(\mathrm{CH}_{2}\right), 70.35$ $(\mathrm{CH}), 80.34\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 114.87,121.62,129.94$ (Ar-ipso), 158.75 (C=O); $m / z(\mathrm{CI}) 268\left([\mathrm{M}+\mathrm{H}]^{+}, 50 \%\right), 212$ (100), 209 (15) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, \quad 268.1551 . \quad \mathrm{C}_{14} \mathrm{H}_{21} \mathrm{NO}_{4}$ requires $\mathrm{m} / \mathrm{z}$, 268.1549).

## rac- N -(tert-Butoxycarbonyl)-2-phenoxymethylaziridine 17

To a solution of 1-( $N$-tert-butoxycarbonylamino)-2-hydroxy-3phenoxypropane rac- $\mathbf{1 6}(0.5 \mathrm{~g}, 1.87 \mathrm{mmol})$ and tosyl chloride
$(0.53 \mathrm{~g}, 2.81 \mathrm{mmol})$ in dry THF ( 10 mL ) was added KOH $(0.52 \mathrm{~g}, 9.35 \mathrm{mmol})$ freshly powered (with the help of a ball mill) at room temperature. The reaction mixture was left stirring overnight. Then it was dissolved in diethyl ether (20 mL ) and filtered. The solvent was removed under reduced pressure and the crude compound was purified using flash chromatography eluting with $10 \%$ ethyl acetate-petroleum ether $40: 60$ to give the aziridine as a pale liquid $(0.31 \mathrm{~g}$, $66.4 \%) . v_{\max }($ neat $) / \mathrm{cm}^{-1}$ 2977, 2931, 2874, 1719, 1599, 1587, 1496, 1457, 1426, 1392, 1367, 1307, 1244, 1222, 1157, 1079, $1038,996,968,854,754,691 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.44(9 \mathrm{H}, \mathrm{s}$, $\left.\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 2.25\left(1 \mathrm{H}, \mathrm{d}, J 4.0, \mathrm{CH} H_{\mathrm{az}}\right), 2.38(1 \mathrm{H}, \mathrm{d}, J 6.0$, $\left.\mathrm{C} H \mathrm{H}_{\mathrm{az}}\right), 2.80-2.86\left(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{\mathrm{az}}\right), 4.0(1 \mathrm{H}, \mathrm{dd}, J 4.9,10.6$, C-CHH-O-Ph), 4.16 ( $1 \mathrm{H}, \mathrm{dd}, J 4.9,10.6, \mathrm{C}-\mathrm{CH} H-\mathrm{O}-\mathrm{Ph}$ ), $6.89-7.33\left(5 \mathrm{H}, \mathrm{m}\right.$, Aryl H); $\delta_{\mathrm{c}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 27.65$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 29.31\left(\mathrm{CH}_{2}\right), 35.57\left(\mathrm{CH}_{2}\right), 67.24(\mathrm{CH}), 81.25$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 114.36,120.92,129.24$ (Ar-ipso), $158.06(\mathrm{C}=\mathrm{O})$; $\mathrm{m} / \mathrm{z}(\mathrm{FAB}) 250\left([\mathrm{M}+\mathrm{H}]^{+}, 80 \%\right), 249$ (70), 194 (100), 176 (15), 154 (80), 136 (34), 107 (29) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 250.1441$. $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{NO}_{3}$ requires $m / z$, 250.1443).

## rac-1-(N-tert-Butoxycarbonylamino)-2-oxo-3-phenoxypropane 18

1-(N-tert-Butoxycarbonylamino)-2-hydroxy-3-phenoxypropane rac- $\mathbf{1 6}(1.77 \mathrm{~g}, 6.62 \mathrm{mmol})$ was added to an ice cold solution of pyridinium dichromate ( $8.51 \mathrm{~g}, 23.17 \mathrm{mmol}$ ) in anhydrous $N, N$ dimethylformamide ( 30 mL ). The reaction mixture was stirred overnight, then worked up with brine $(40 \mathrm{~mL})$. The compound was extracted with diethyl ether $(3 \times 40 \mathrm{~mL})$. The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified by flash chromatography eluting with $10 \%$ ethyl acetate-petroleum ether $40: 60$ to give 1-( $N$-tert-butoxycarbonylamino)-2-oxo-3-phenoxypropane 18 as a white solid $(0.54 \mathrm{~g}, 31.0 \%), \mathrm{mp} 65^{\circ} \mathrm{C}$; $v_{\max }($ neat $) / \mathrm{cm}^{-1} 3438$, 2983, 2254, 1711, 1600, 1496, 1369, 1246, 1170, 1066, 909, 732; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.46\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 4.34(2 \mathrm{H}, \mathrm{d}, J 5.1$, $\left.\mathrm{CH}_{2}-\mathrm{NH}\right)$, $4.64\left(2 \mathrm{H}, \mathrm{s}, \mathrm{O}-\mathrm{CH}_{2}-\mathrm{C}(=\mathrm{O})\right.$ ), $5.23(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH})$, $6.86-7.35\left(5 \mathrm{H}, \mathrm{m}\right.$, Aryl-H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.69$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 48.81 \quad\left(\mathrm{CH}_{2}\right), \quad 72.16 \quad\left(\mathrm{CH}_{2}\right), \quad 80.46 \quad\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right)$, 114.83, 122.43, 130.18 (Ar-ipso), 157.82 (C=O), 164.50 (C=O); $\mathrm{m} / \mathrm{z}(\mathrm{CI}) 266\left([\mathrm{M}+\mathrm{H}]^{+}, 87 \%\right), 265(100), 258$ (10), 242 (7), 239 (5) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 266.1392. $\mathrm{C}_{14} \mathrm{H}_{19} \mathrm{NO}_{4}$ requires $\mathrm{m} / \mathrm{z}$, 266.1392).

## $R$-(-)-1-( $N$-tert-Butoxycarbonylamino)-2-hydroxy-3-phenoxypropane 16

A mixture of ( $p$-cymene) ruthenium(II) chloride dimer ( 0.31 mg , $0.0051 \mathrm{mmol})$ and $(1 R, 2 R)$-TsDPEN $2(0.37 \mathrm{mg}, 0.0102 \mathrm{mmol})$ in a $5: 2$ formic acid-triethylamine mixture $(2.5 \mathrm{~mL})$ was stirred at $28^{\circ} \mathrm{C}$ for 15 min . 1-( $N$-tert-Butoxycarbonylamino)-2-oxo-3phenoxypropane $18(0.54 \mathrm{~g}, 2.04 \mathrm{mmol})$ was added and the solution was stirred at $28^{\circ} \mathrm{C}$ for 24 hours. Then the mixture was filtered through silica and washed with ethyl acetate ( 60 ml ). The solvent was evaporated under reduced pressure to give the crude compound, which was purified by flash chromatography ( $20 \% \mathrm{v} / \mathrm{v}$ ethyl acetate-petroleum ether $40: 60$ ) to give the $R$ -$(-)-16$ as a white solid $(0.47 \mathrm{~g}, 86.2 \%)$. The product was determined to be of $84 \%$ ee by HPLC analysis (Chiral OD, hexane-ethanol-diethylamine $=95: 5: 0.1\left(0.5 \mathrm{~mL} \mathrm{~min}{ }^{-1}\right), R$ isomer $22.48 \mathrm{~min}, S$ isomer 41.64 min ), $\mathrm{mp} 70-72^{\circ} \mathrm{C}$; $[a]_{\mathrm{D}}^{20}-7.2(c=1$, ethanol); $v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 3359,3060,2977,2931,1690,1599$, 1587, 1497, 1456, 1391, 1366, 1245, 1170, 1123, 1070, 1044, 993, $925,855,813,781,754,737,691 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.46$ ( $\left.9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 3.26-3.35(2 \mathrm{H}, \mathrm{m}, \mathrm{CHH}, \mathrm{C}(\mathrm{OH}) \mathrm{H}), 3.44-3.52$ ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CHH}$ ), $3.90-4.01\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2}\right), 4.11(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH})$, $4.99(1 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{NH}), 6.88-7.33\left(5 \mathrm{H}, \mathrm{m}\right.$, Aryl-H); $\delta_{\mathrm{C}}(75.5 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 28.74\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 43.89\left(\mathrm{CH}_{2}\right), 69.62\left(\mathrm{CH}_{2}\right), 70.39(\mathrm{CH})$, $80.34\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 114.87,121.64,129.95$ (Ar-ipso), 158.74
$(\mathrm{C}=\mathrm{O}) ; m / z(\mathrm{CI}) 268\left([\mathrm{M}+\mathrm{H}]^{+}, 35 \%\right), 212(63), 192(11)$, 168 (35), 154 (100), 136 (70), 107 (26) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 268.155464. $\mathrm{C}_{14} \mathrm{H}_{21} \mathrm{NO}_{4}$ requires $m / z, 268.154883$ ). The reduction using $(1 S, 2 R)$-1-propan-2-ol was attempted using identical conditions to those reported however no reduction product was isolated. ${ }^{4 b}$

## (R)-(-)-N-(tert-Butoxycarbonyl)-2-phenoxymethylaziridine 17

To a solution of chiral 1- $N$-(tert-butoxycarbonylamino)-2-hydroxy-3-phenoxypropane $R$-(-)-16 ( $0.38 \mathrm{~g}, 1.42 \mathrm{mmol}$ ) and tosyl chloride ( $0.41 \mathrm{~g}, 2.13 \mathrm{mmol}$ ) in dry THF ( 10 mL ) was added $\mathrm{KOH}(0.40 \mathrm{~g}, 7.10 \mathrm{mmol}$ ) freshly powdered (with the help of a ball mill) at room temperature. The reaction mixture was left stirring overnight. Then it was dissolved in diethyl ether $(20 \mathrm{~mL})$ and filtered. The solvent was removed under reduced pressure and the crude compound was purified using flash chromatography eluting with $20 \%$ diethyl ether-hexane to give the aziridine $R-(-)-17$ as a colourless liquid ( $0.22 \mathrm{~g}, 62.9 \%$ ). The product was determined to be of $83.7 \%$ ee by HPLC analysis (Chiral OD, hexane-ethanol-diethylamine $=95: 5: 0.1(0.5$ $\mathrm{mL} \mathrm{min}{ }^{-1}$ ), $R$ isomer $22.48 \mathrm{~min}, S$ isomer 41.64 min ); $[a]_{D}^{20}$ -68.9 ( $c=1$, ethanol); $v_{\max }($ neat $) / \mathrm{cm}^{-1} 2979,2932,2360,1722$, 1600, 1497, 1458, 1393, 1308, 1245, 1222, 1156, 1110, 1080, $1039,996,968,853,832,754,692 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.44$ $\left(9 \mathrm{H}, \mathrm{s},\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 2.25\left(1 \mathrm{H}, \mathrm{d}, J 3.6, \mathrm{CH} H_{\mathrm{az}}\right), 2.38(1 \mathrm{H}, \mathrm{d}$, $\left.J 6.0, \mathrm{C}_{\mathrm{Hz}}\right), 2.80-2.87\left(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H_{\mathrm{az}}\right), 4.0(1 \mathrm{H}, \mathrm{dd}, J 4.9$, 10.6, C-CHH-O-Ph), 4.16 ( 1 H , dd, $J 4.9,10.6$, C-CHH-O-Ph), 6.89-7.31 ( $5 \mathrm{H}, \mathrm{m}$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 27.65$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 29.32\left(\mathrm{CH}_{2}\right), 35.57\left(\mathrm{CH}_{2}\right), 67.25(\mathrm{CH}), 81.26$ $\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 114.36,120.92,129.24$ (Ar-ipso), $158.05(\mathrm{C}=\mathrm{O}) ; \mathrm{m} / \mathrm{z}$ (FAB) 250, ([M] ${ }^{+}, 80 \%$ ), 249 (70), 194 (100), 176 (15), 154 (80), 136 (34), 107 (29) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 250.1441 . \mathrm{C}_{14} \mathrm{H}_{19} \mathrm{NO}_{3}$ requires $m / z, 250.1443$ ).

## 2-(Chloromethyl)allyl phenyl ether

Sodium hydride ( $0.10 \mathrm{~g}, 2.5 \mathrm{mmol}$ ) was stirred in DMF ( 7 mL ) at $0^{\circ} \mathrm{C}$ and to this phenol $(0.094 \mathrm{~g}, 1 \mathrm{mmol})$ was added slowly. Once the addition was complete the reaction was stirred for 20 minutes. The cooling bath was removed and the stirring was continued for a further hour at room temperature. 3-Chloro-2-chloromethylprop-1-ene ( $0.125 \mathrm{~g}, 1 \mathrm{mmol}$ ) was added and the reaction was stirred overnight. Saturated ammonium chloride solution ( 10 mL ) was added to quench the reaction and then the aqueous solution was extracted with ethyl acetate. The organic layers were dried over magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified using medium pressure flash chromatography eluting with $1 \%$ ether-hexane to give the product as a pale oil ( $0.08 \mathrm{~g}, 43.7 \%$ ). $v_{\text {max }}$ (neat) $/ \mathrm{cm}^{-1} 3062,3039,2925,2867$, 1930, 1840, 1775, 1702, 1654, 1597, 1495, 1456, 1443, 1407 , 1368, 1334, 1299, 1241, 1172, 1079, 1056, 1033, 993, 919, 883, $814,753,691,665 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 4.19(2 \mathrm{H}, \mathrm{br} \mathrm{d}, J 0.7$, $\left.\mathrm{CH}_{2}\right), 4.63\left(2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{CH}_{2}\right), 5.37(1 \mathrm{H}, \mathrm{m},=\mathrm{CHH}), 5.39(1 \mathrm{H}, \mathrm{m}$, $=\mathrm{CH} H), 6.90-7.00(3 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 7.25-7.35(2 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H})$; $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 45.52\left(\mathrm{CH}_{2}\right), 68.26\left(\mathrm{CH}_{2}\right), 115.17$, 117.93, 121.56, 129.91, 141.25 (Ar-ipso), $158.80\left(C=\mathrm{CH}_{2}\right)$, $164.35\left(\mathrm{C}=\mathrm{CH}_{2}\right) ; m / z(\mathrm{EI}) 184,182\left([\mathrm{M}+\mathrm{H}]^{+} 62,86 \%\right), 148$, 147 (57, 100), 133, $131(50,26), 119,118(15,7), 94,93(36,6)$ (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 184.0472,182.0501 . \mathrm{C}_{14} \mathrm{H}_{19} \mathrm{NO}_{4}$ requires $m / z, 184.0469,182.0498)$.

## $N$-(tert-Butoxycarbonyl)- N -(benzyl)-2-(phenoxymethyl)allylamine

Sodium hydride ( $0.62 \mathrm{~g}, 5.80 \mathrm{mmol}$ ) was stirred in DMF ( 20 mL ) and to this benzylamine ( $0.62 \mathrm{~g}, 5.80 \mathrm{mmol}$ ) in 5 mL of DMF was added slowly. Once the addition was completed the reaction was stirred for a further 1 hour at room temperature. Then 2-(chloromethyl)allyl phenyl ether in 5 mL of DMF
was added and the reaction mixture was stirred overnight. The reaction was then filtered and DMF was removed. Dichloromethane ( 30 mL ) was then added followed by di-tert-butyl dicarbonate $(1.27 \mathrm{~g}, 5.80 \mathrm{mmol})$ dissolved in 10 mL of dichloromethane. The reaction mixture was stirred overnight, then saturated ammonium chloride solution ( 30 mL ) added. The phases were separated and the aqueous layer extracted with dichloromethane $(2 \times 30 \mathrm{~mL})$. The combined organic layers were dried over magnesium sulfate, filtered, and the solvent removed in vacuo. Purification by flash chromatography on silica with $10 \%$ ethyl acetate-hexane $60: 80$, gave the product as a colourless liquid ( $0.56 \mathrm{~g}, 27.3 \%$ ); $v_{\max }($ neat $) / \mathrm{cm}^{-1} 3366,3064$, 3030, 2976, 2929, 1950, 1809, 1694, 1599, 1587, 1495, 1454, 1409, 1365, 1339, 1301, 1243, 1169, 1119, 1078, 1053, 1031, $993,915,882,816,754,692 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.45(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 3.91\left(2 \mathrm{H}, \mathrm{m}, \mathrm{Ph}-\mathrm{O}-\mathrm{CH}_{2}\right), 4.45\left(4 \mathrm{H}, \mathrm{br} \mathrm{m},=\mathrm{C}-\mathrm{CH}_{2}\right.$, $\left.\mathrm{CH}_{2}-\mathrm{Ph}\right), 5.04(1 \mathrm{H}, \mathrm{s},=\mathrm{CHH}), 5.11(1 \mathrm{H}, \mathrm{br} \mathrm{s},=\mathrm{CHH}), 6.88-$ $7.38(10 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.76\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right)$, $48.63\left(\mathrm{CH}_{2}\right), 49.40\left(\mathrm{CH}_{2}\right), 50.07\left(\mathrm{CH}_{2}\right), 60.15\left(\mathrm{C}=\mathrm{CH}_{2}\right), 71.27$ $\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 80.49\left(\mathrm{C}=\mathrm{CH}_{2}\right), 114.53,115.07,121.32,127.65$, 127.87, 128.38, 128.92, 129.83, 138.38, 140.91 (Ar-ipso), 158.89 (C=O); $m / z$ (FAB) 354 ([M] ${ }^{+}, 70 \%$ ), 307 (25), 298 (87), 254 (70), 204 (90), 160 (23), 154 (90), 136 (62), 107 (27) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 354.2064 . \mathrm{C}_{22} \mathrm{H}_{27} \mathrm{NO}_{3}$ requires $m / z, 354.2069$ ).

## 1-( $N$-Butoxycarbonyl- $N$-benzylamino)-2-oxo-3-phenoxypropane 20

$N$-(tert-Butoxycarbonyl)- $N$-(benzyl)-2-(phenoxymethyl)allylamine ( $0.17 \mathrm{~g}, 0.48 \mathrm{mmol}$ ) was stirred in dichloromethane ( 10 mL ) at $-78^{\circ} \mathrm{C}$. An empty trap followed by a trap containing a solution of $5 \%$ potassium iodide in $50 \%$ acetic acid-water were connected to the outlet. Ozone was passed through the reaction mixture for 20 minutes after which time the solution appeared pale blue. Oxygen was bubbled through the solution for 10 minutes followed by nitrogen for 20 minutes. Triphenylphosphine ( $0.19 \mathrm{~g}, 0.72 \mathrm{mmol}$ ) was added and the cooling bath removed. The reaction was stirred overnight, the solvent removed and purified by column chromatography yielding the product 20 as a colourless liquid of high viscosity ( $0.08 \mathrm{~g}, 47 \%$ yield). $v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 3371,3064,2977,2930,1740,1694,1599$, 1589, 1495, 1454, 1427, 1393, 1366, 1292, 1245, 1165, 1126, $1080,1060,967,886,754,692 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}, 1: 1\right.$ mixture of NCO rotamers) $1.44\left(4.5 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.49(4.5 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 4.12(1 \mathrm{H}, \mathrm{s}, \mathrm{N}-\mathrm{CHH}-\mathrm{Ph}), 4.20(1 \mathrm{H}, \mathrm{s}, \mathrm{N}-\mathrm{CHH}-\mathrm{Ph})$, $4.48(1 \mathrm{H}, \mathrm{s}, \mathrm{C}(=\mathrm{O})-\mathrm{CHH}), 4.51(2 \mathrm{H}, \mathrm{s}$, Ph-O-CH2$), 4.59(1 \mathrm{H}$, $\mathrm{s}, \mathrm{C}(=\mathrm{O})-\mathrm{CHH}), 6.80-7.35(10 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}) ; \delta_{\mathrm{C}}(75.5 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 28.64\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 51.60\left(\mathrm{CH}_{2}\right), 53.70\left(\mathrm{CH}_{2}\right), 72.24$ $\left(\mathrm{CH}_{2}\right), 81.06\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right), 80.49\left(\mathrm{C}=\mathrm{CH}_{2}\right), 114.78,122.18,127.98 \text {, }}\right.$ 128.59, 129.05, 130.07, 137.88 (Ar-ipso), 156.21 (C=O), 157.97 (C=O); m/z (FAB) 356 ([M + H] ${ }^{+}, 31 \%$ ), 307 (20), 300 (100), 289 (10), 256 (26), 220 (21), 154 (69), 136 (45), 120 (32) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 356.1862 . \mathrm{C}_{21} \mathrm{H}_{25} \mathrm{NO}_{4}$ requires $m / z, 356.1862$ ).

## $R$-(+)-1-( $N$-tert-Butoxycarbonyl- $N$-benzylamino)-2-hydroxy-3phenoxypropane 21

A mixture of ( $p$-cymene)ruthenium(II) chloride dimer ( 0.26 mg , $0.0043 \mathrm{mmol})$ and $(1 R, 2 R)$-TsDPEN $2(0.30 \mathrm{mg}, 0.0086 \mathrm{mmol})$ in a $5: 2$ formic acid-triethylamine mixture $(2.5 \mathrm{~mL})$ was stirred at $28^{\circ} \mathrm{C}$ for 15 min . 1-( $N$-tert-Butoxycarbonyl- $N$-benzylamino)-2-oxo-3-phenoxypropane ( $0.61 \mathrm{~g}, 1.72 \mathrm{mmol}$ ) was added and the solution was stirred at $28^{\circ} \mathrm{C}$ for 24 hours. Then the mixture was filtered through silica and washed with ethyl acetate ( 60 ml ). The solvent was evaporated under reduced pressure to give the crude compound, which was purified by flash chromatography ( $20 \% \mathrm{v} / \mathrm{v}$ ethyl acetate-petroleum ether $40: 60$ ) to give the product 21 as a colourless liquid $(0.61 \mathrm{~g}, 99.0 \%)$. The product was determined to be of $59.7 \%$ ee by HPLC analysis (Chiral OD, hexane-ethanol-diethylamine $=95: 5: 0.1 \quad(0.5 \mathrm{~mL}$ $\min ^{-1}$ ), $S$ isomer $21.09 \mathrm{~min}, R$ isomer 22.98 min$)$; $[a]_{\mathrm{D}}^{20}+7.45$
( $c=2$, ethanol); $v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 3430,3061,3028,2973,2928$, 1947, 1736, 1694, 1598, 1586, 1495, 1453, 1413, 1391, 1365, $1288,1244,1168,1132,1077,995,967,924,881,814,754,691 ;$ $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.47\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 3.44-3.53(2 \mathrm{H}, \mathrm{m}$, $\mathrm{C}(\mathrm{OH}) \mathrm{H}, \mathrm{OH}), 3.82-3.94\left(2 \mathrm{H}, \mathrm{m}\right.$, Boc- $\left.\mathrm{N}-\mathrm{CH}_{2}-\mathrm{Ph}\right), 4.06-4.23$ $\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}(\mathrm{OH})-\mathrm{C} \mathrm{H}_{2}\right.$-N-Boc), $4.48\left(2 \mathrm{H}, \mathrm{AB}\right.$ system, $J^{\mathrm{AB}} 14.7$, $\left.\mathrm{Ph}-\mathrm{O}-\mathrm{CH}_{2}\right), 6.87-7.36\left(10 \mathrm{H}, \mathrm{m}\right.$, Aryl- $H$ ); $\delta_{\mathrm{C}}(75.5 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 28.15\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 50.41\left(\mathrm{CH}_{2}\right), 52.40\left(\mathrm{CH}_{2}\right), 68.78$ $\left(\mathrm{CH}_{2}\right), 69.98(\mathrm{CH}), 81.00\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 114.21,120.86,127.16$, 128.39, 129.31, 137.74 (Ar-ipso), 164.09 (C=O); $m / z$ (FAB) 358 ( $\left.[\mathrm{M}+\mathrm{H}]^{+}, 71 \%\right), 302$ (90), 282 (15), 258 (91), 208 (35), 194 (10), 154 (86), 164 (16), 136 (60), 120 (44) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 358.2014. $\mathrm{C}_{21} \mathrm{H}_{27} \mathrm{NO}_{4}$ requires $m / z, 358.2018$ ). The reduction using ( $1 S, 2 R$ )-1-propan- 2 -ol was carried out using identical conditions to those reported. ${ }^{4 b}$ The product, $R-(+)-21$ was formed in $99 \%$ yield and $61 \%$ ee as determined by chiral HPLC analysis.

## 1-( $N$-tert-Butoxycarbonyl- N -benzylamino)-2-hydroxy-3-phenoxypropane rac-21

1-( $N$-tert-Butoxycarbonyl- $N$-benzylamino)-2-oxo-3-phenoxypropane ( $0.70 \mathrm{~g}, 1.97 \mathrm{mmol}$ ) was dissolved in 15 mL of $\mathrm{MeOH}-$ $\mathrm{H}_{2} \mathrm{O}(9: 1)$, in an ice bath. Sodium borohydride ( $0.25 \mathrm{~g}, 7.00$ mmol ) dissolved in 10 mL of $\mathrm{MeOH}-\mathrm{H}_{2} \mathrm{O}(9: 1)$ was added slowly to the ketone solution. The reaction mixture was allowed to warm up to room temperature. When the reaction was completed (TLC), it was concentrated under reduced pressure. The residue was dissolved in diethyl ether ( 30 mL ) and worked up with saturated ammonium chloride solution ( 30 mL ). The phases were separated and the aqueous layer extracted with diethyl ether $(2 \times 20 \mathrm{~mL})$. The organic layers were dried over anhydrous magnesium sulfate, filtered and the solvent removed under reduced pressure. The crude compound was purified using medium pressure flash chromatography eluting with $20 \%$ ethyl acetate-petroleum ether $40: 60$ to give 1-( $N$-tert-butoxycarbonyl- $N$-benzylamino)-2-hydroxy-3-phenoxypropane 21 as a colourless liquid ( $0.70 \mathrm{~g}, 99 \%$ ); $v_{\max }($ neat $) / \mathrm{cm}^{-1} 3425$, 3061, 3028, 2973, 2928, 1948, 1693, 1598, 1586, 1537, 1494, 1453, 1413, 1365, 1288, 1244, 1169, 1133, 1077, 1043, 995, 966, 923, 881, 814, 753, 691; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.47(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 3.44-3.57(2 \mathrm{H}, \mathrm{m}, \mathrm{C}(\mathrm{OH}) \mathrm{H}, \mathrm{OH}), 3.88-3.94(2 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{N}(\mathrm{Boc})-\mathrm{CH}_{2}-\mathrm{Ph}\right), 4.06-4.17\left(2 \mathrm{H}, \mathrm{m}, \mathrm{CH}(\mathrm{OH}) \mathrm{CH}_{2}\right), 4.48(2 \mathrm{H}$, AB system, $\left.J^{\mathrm{AB}} 14.7, \mathrm{Ph}-\mathrm{O}-\mathrm{CH}_{2}\right), 6.87-7.36(10 \mathrm{H}, \mathrm{m}$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 28.77\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 51.02\left(\mathrm{CH}_{2}\right), 53.02$ $\left(\mathrm{CH}_{2}\right), 69.42\left(\mathrm{CH}_{2}\right), 70.61(\mathrm{CH}), 86.50\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 114.82$, 121.46, 127.78, 129.01, 129.93, 137.50 (Ar-ipso), 164.14 (C=O); $\mathrm{m} / \mathrm{z}$ (FAB) 358 ( $[\mathrm{M}+\mathrm{H}]^{+}, 31 \%$ ), 307 (27), 302 (40), 289 (16), 258 (43), 208 (17), 154 (100), 136 (72), 120 (30) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 358.2021 . \mathrm{C}_{21} \mathrm{H}_{27} \mathrm{NO}_{4}$ requires $m / z, 358.2018$ ).

## $\boldsymbol{R}-(+)-1-(\boldsymbol{N}$-Benzylamino)-2-hydroxy-3-phenoxypropane

To a solution of 1-( $N$-tert-butoxycarbonyl- $N$-benzylamino)-2-hydroxy-3-phenoxypropane $R-(+)-21(0.48 \mathrm{~g}, 1.34 \mathrm{mmol})$ in dichloromethane ( 2 mL ) TFA ( 2 mL ) was added slowly. After 2 hours the solvent was removed and the resulting product was dissolved in ethyl acetate and 0.2 M NaOH solution was added until the $\mathrm{pH}=7$. The product was then extracted with ethyl acetate and the combined organic layers were dried over magnesium sulfate, filtered and the solvent removed in vacuo. Recrystallisation (ethyl acetate-petroleum ether $40: 60$ ) gave 1-( $N$-benzylamino)-2-hydroxy-3-phenoxypropane as a white solid ( $0.34 \mathrm{~g}, 47.0 \%$ yield). The product was determined to be of $45.6 \%$ ee by HPLC analysis (Chiral OD, hexane-ethanoldiethylamine $=95: 5: 0.1\left(0.5 \mathrm{~mL} \mathrm{~min}^{-1}\right), S$ isomer 21.27 min , $R$ isomer 60.49 min ), $\mathrm{mp} 64-66^{\circ} \mathrm{C} ;[a]_{\mathrm{D}}^{20}+9.4(c=2$, ethanol); $v_{\max }\left(\mathrm{CDCl}_{3}\right) / \mathrm{cm}^{-1} 3375,2926,2252,1681,1599,1495,1454$, $1244,1043,908,732,649 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.24(2 \mathrm{H}, \mathrm{br} \mathrm{s}$, $\mathrm{OH}, \mathrm{NH}), 2.81\left(1 \mathrm{H}, \mathrm{dd}, J^{\mathrm{b}} 12.1, J^{\mathrm{bc}} 7.7, \mathrm{C} H^{\mathrm{b}} \mathrm{H}^{\mathrm{a}} \mathrm{CH}^{\mathrm{c}}(\mathrm{OH})\right)$, $2.91\left(1 \mathrm{H}, \mathrm{dd}, J^{\mathrm{ab}} 12.2, J^{\mathrm{ac}} 3.9, \mathrm{CH}^{\mathrm{b}} H^{\mathrm{a}} \mathrm{CH}^{\mathrm{c}}(\mathrm{OH})\right), 3.85(2 \mathrm{H}$,
$\left.\mathrm{AB}_{\text {system }}, J^{\mathrm{AB}} 13.0, \mathrm{NHCH}_{2}-\mathrm{Ph}\right), 3.98\left(2 \mathrm{H}, \mathrm{d}, J 5.3, \mathrm{Ph}(\mathrm{O})-\mathrm{CH}_{2}\right)$, 4.05-4.13 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CH}(\mathrm{OH})$ ), 6.89-7.35 ( $10 \mathrm{H}, \mathrm{m}$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 51.48\left(\mathrm{CH}_{2}\right), 54.10\left(\mathrm{CH}_{2}\right), 68.66(\mathrm{CH})$, $70.64\left(\mathrm{CH}_{2}\right), 114.92,121.48,127.69,128.59,128.95,129.89$ (Ar-ipso); $m / z$ (CI) 258 ([M + H] ${ }^{+}, 80 \%$ ), 240 (5), 148 (7), 120 (18), 108 (8), 91 (10) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 258.1498 . \mathrm{C}_{16} \mathrm{H}_{19} \mathrm{NO}_{2}$ requires $m / z, 258.1494)$.

## rac-1-( $N$-Benzylamino)-2-hydroxy-3-phenoxypropane

To a solution of 1-( $N$-tert-butoxycarbonyl- $N$-benzylamino)-2-hydroxy-3-phenoxypropane rac-21 ( $0.76 \mathrm{~g}, 2.11 \mathrm{mmol}$ ) in dichloromethane ( 2 mL ) TFA ( 2 mL ) was added slowly. After 2 hours the solvent was removed and the resulting product was dissolved in ethyl acetate and NaOH solution 0.2 M was added until the $\mathrm{pH}=7$. The product was then extracted with ethyl acetate and the combined organic layers were dried over magnesium sulfate, filtered and the solvent removed in vacuo. Recrystallisation (ethyl acetate-petroleum ether $40: 60$ ) gave 1-( $N$-benzylamino)-2-hydroxy-3-phenoxypropane (racemic) as a white solid ( $0.34 \mathrm{~g}, 47 \%$ yield). Mp $74-76^{\circ} \mathrm{C}$; $v_{\text {max }}\left(\mathrm{CDCl}_{3}\right) /$ $\mathrm{cm}^{-1} 3402,2925,1644,1599,1587,1453,1301,1245,1173$, $1078,1041,909,752,691 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.55(2 \mathrm{H}, \mathrm{br} \mathrm{s}$, $\mathrm{OH}, \mathrm{NH}), 2.78\left(1 \mathrm{H}, \mathrm{dd}, J^{\mathrm{ab}} 12.06, J^{\mathrm{bc}} 7.72, \mathrm{C}^{\mathrm{b}} \mathrm{H}^{\mathrm{a}} \mathrm{CH}^{\mathrm{c}}(\mathrm{OH})\right.$ ), $2.88\left(1 \mathrm{H}, \mathrm{dd}, J^{\mathrm{ab}} 12.25, J^{\mathrm{ac}} 3.96, \mathrm{CH}^{\mathrm{b}} H^{\mathrm{a}} \mathrm{CH}^{\mathrm{c}}(\mathrm{OH})\right), 3.83(2 \mathrm{H}$, $\left.\mathrm{AB}_{\text {system }}, J^{\mathrm{AB}} 13.37, \mathrm{NHCH}_{2}-\mathrm{Ph}\right), 3.96(2 \mathrm{H}, \mathrm{d}, J 5.09, \mathrm{Ph}(\mathrm{O})-$ $\left.\mathrm{CH}_{2}\right)$, 4.04-4.15 $(1 \mathrm{H}, \mathrm{m}, \mathrm{CH}(\mathrm{OH}))$, 6.84-7.38 $(10 \mathrm{H}, \mathrm{m}$, Aryl $\mathrm{H}) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 51.48\left(\mathrm{CH}_{2}\right), 54.10\left(\mathrm{CH}_{2}\right), 68.66$ $(\mathrm{CH}), 70.64\left(\mathrm{CH}_{2}\right), 114.93,121.44,127.58,128.55,128.92$, 129.45, 129.89, 140.24 (Ar-ipso); m/z (CI) 258 ( $[\mathrm{M}+\mathrm{H}]^{+}, 90 \%$ ), 256 (17), 168 (21), 164 (18), 162 (12), 146 (8), 120 (31), 106 (10) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 258.1498 . \mathrm{C}_{16} \mathrm{H}_{19} \mathrm{NO}_{2}$ requires $\mathrm{m} / \mathrm{z}$, 258.1494).

O-(tert-Butyldimethylsilyl)-1-(N-tert-butoxycarbonylamino)-2-hydroxy-3-phenoxypropane and subsequently $R-(+)-19$ from $R-(+)-16$
1-(N-tert-Butoxycarbonylamino)-2-hydroxy-3-phenoxypropane $R-(-)-16(0.3 \mathrm{~g}, 1.12 \mathrm{mmol})$ and imidazole $(0.19 \mathrm{~g}, 2.80 \mathrm{mmol})$ were dissolved in 10 mL of DMF and treated with tertbutylchlorodimethylsilane (TBDMSCl) ( $0.21 \mathrm{~g}, 1.35 \mathrm{mmol}$ ). The mixture was stirred at $35-45^{\circ} \mathrm{C}$ until the starting materials were completely consumed (followed by TLC). The reaction was then treated with sodium bicarbonate saturated solution $(20 \mathrm{~mL})$ and extracted with diethyl ether $(3 \times 20 \mathrm{~mL})$. The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified by flash chromatography eluting with $20 \%$ ethyl acetate-petroleum ether $40: 60$ to give $O$-(tert-butyldimethylsilyl)-1-(N-tert-butoxy-carbonylamino)-2-hydroxy-3-phenoxypropane as a colourless liquid ( $0.42 \mathrm{~g}, 99 \%$ ); $[a]_{\mathrm{D}}^{20}-12.05\left(c=2\right.$, ethanol); $v_{\max }($ neat $) /$ $\mathrm{cm}^{-1} 2929,1703,1674,1652,1600,1497,1390,1366,1328$, $1245,1172,1062,987,937,836,777,752,691 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right)-0.0001\left(3 \mathrm{H}, \mathrm{s}, \operatorname{SiCH} H_{3}\left(\mathrm{CH}_{3}\right)_{3}\right), 0.0231(3 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)_{3}\right), 0.80\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.33\left(9 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right)$, 3.08-3.16 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CH} H-\mathrm{NH}), 3.26-3.34(1 \mathrm{H}, \mathrm{m}, \mathrm{C} H \mathrm{H}-\mathrm{NH})$, 3.76 ( 2 H , dd, $J$ 1.32, 5.75, Ph-O-CH2), 3.98-4.04 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{CH}$ ), $4.70(1 \mathrm{H}, \mathrm{brs}, \mathrm{N} H), 6.75-7.25(5 \mathrm{H}, \mathrm{m}$, $\operatorname{Aryl} \mathrm{H}) ; \delta_{\mathrm{C}}(75.5 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right)$ - $4.46\left(\mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)_{3}\right)$, $-4.11\left(\mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)_{3}\right), 26.20(\mathrm{Si}-$ $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 28.79\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right), 44.65\left(\mathrm{CH}_{2}\right), 70.31\left(\mathrm{CH}_{2}\right), 70.40$ (CH), 114.78, 121.29, 122.29, 129.86, 135.54 (Ar-ipso), 163.97 (C=O); $m / z$ (FAB) $382\left([\mathrm{M}+\mathrm{H}]^{+}, 25 \%\right), 326$ (26), 282 (63), 268 (100), 232 (6), 174 (8), 154 (20), 136 (16), 107 (10) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 382.2413 . \mathrm{C}_{20} \mathrm{H}_{35} \mathrm{NSiO}_{4}$ requires $\mathrm{m} / \mathrm{z}, 382.2413$ ). This was converted into $R-(+)-19$ ( 100 mg scale) by treatment with sodium hydride-benzyl bromide using the same procedure as employed for $S-(+)-6$ previously described. The product was formed in $50 \%$ yield and exhibited $[a]_{D}^{20}+0.70(c=2$, ethanol).

## $O$-(tert-Butyldimethylsily)-1-( $N$-tert-butoxycarbonyl- $N$-benzyl-

 amino)-2-hydroxy-3-phenoxypropane $S$-(+)-19 from $R$-(+)-21$R$-(+)-1-( $N$-tert-Butoxycarbonyl- $N$-benzylamino)-2-hydroxy-3-phenoxypropane $21(0.2 \mathrm{~g}, 0.56 \mathrm{mmol})$ and imidazole ( 0.08 g , 1.12 mmol ) were dissolved in 10 mL of DMF and treated with tert-butylchlorodimethylsilane (TBDMSCl) $(0.13 \mathrm{~g}, 0.84$ mmol ). The mixture was stirred at $35-45^{\circ} \mathrm{C}$ until the starting materials had been completely consumed (followed by TLC). The reaction was then treated with sodium bicarbonate saturated solution $(20 \mathrm{~mL})$ and extracted with diethyl ether ( $3 \times 20$ mL ). The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified by flash chromatography eluting with $20 \%$ ethyl acetate-petroleum ether 40 : 60 to give $O$-(tert-butyldimethylsilyl-1-( $N$-tert-butoxy-carbonyl- $N$-benzylamino)-2-hydroxy-3-phenoxypropane $\quad R$ -$(+)-19$ as a colourless liquid $(0.25 \mathrm{~g}, 99 \%) ;[a]_{\mathrm{D}}^{20}+0.7(c=2$, ethanol); $v_{\max }($ neat $) / \mathrm{cm}^{-1} 3429,3064,3031,2955,2929,2857$, 1696, 1600, 1587, 1496, 1462, 1409, 1366, 1245, 1171, 1135, $1078,1049,980,880,836,808,775,753,691 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right)-0.10\left(6 \mathrm{H}, \mathrm{s}, \mathrm{Si}\left(\mathrm{CH}_{3}\right)_{2}\right), 0.90\left(9 \mathrm{H}, \mathrm{s}, \mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.44$ $\left(4.5 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 1.51\left(4.5 \mathrm{H}, \mathrm{s}, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 3.13-3.20(1 \mathrm{H}, \mathrm{m}$, $\mathrm{C}(\mathrm{O}) \mathrm{Si}-\mathrm{CH} H-\mathrm{NBoc}), 3.44-3.59(1 \mathrm{H}, \mathrm{m}, \mathrm{C}(\mathrm{O}) \mathrm{SiCHH}-\mathrm{NBoc})$, 3.80-3.93 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{N}-\mathrm{CH}_{2}-\mathrm{Ph}$ ), 4.40 ( $2 \mathrm{H}, \mathrm{d}, J 15.45$, $\mathrm{Ph}-\mathrm{O}-$ $\mathrm{CH}_{2}$ ), 4.70-4.74 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{Ph}-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}(\mathrm{OSi})$ ), 6.80-7.30 (10 $\mathrm{H}, \mathrm{m}$, Aryl H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)-4.33\left(\mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)_{3}\right)$, -4.14 $\left(\mathrm{SiCH}_{3}\left(\mathrm{CH}_{3}\right)_{3}\right), 18.48\left(\mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 26.27\left(\left(\mathrm{CH}_{3}\right)_{3} \mathrm{C}\right)$, $28.81\left(\mathrm{Si}-\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 50.26,50.83\left(\mathrm{CH}_{2}\right), 51.95,52.92\left(\mathrm{CH}_{2}\right)$, $69.31\left(\mathrm{CH}_{2}\right), 70.26,70.64(\mathrm{CH}), 77.42,77.85\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 105.42$, 114.78, 117.77, 120.77, 121.07, 122.48, 125.57, 126.18, 126.78, 127.55, 127.84, 128.14, 128.91, 129.82, 133.73, 163.95 (C=O); $\mathrm{m} / \mathrm{z}(\mathrm{FAB}) 472$ ( $\left.[\mathrm{M}+\mathrm{H}]^{+}, 27 \%\right), 372$ (100), 358 (95), 322 (7), 154 (14), 136 (12), 120 (17) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 472.2883$. $\mathrm{C}_{27} \mathrm{H}_{41} \mathrm{NSiO}_{4}$ requires $m / z, 472.2883$ ).

## ( $\pm$ )-3-(1-Naphthyloxy)-1,2-epoxypropane

To a solution of 1-naphthol ( $2 \mathrm{~g}, 13.87 \mathrm{mmol}$ ) in 20 mL of dry DMF was added $\mathrm{NaH}(1.11 \mathrm{~g}, 27.74 \mathrm{mmol})$ and the mixture was heated to $80^{\circ} \mathrm{C}$ for 1 hour. To this solution was added epichlorohydrin ( $2.57 \mathrm{~g}, 27.74 \mathrm{mmol}$ ) and it was stirred for 4 hours at $80^{\circ} \mathrm{C}$. The solution was cooled and poured into 30 mL of water. The product was then extracted with diethyl ether $(3 \times 30 \mathrm{~mL})$, dried with $\mathrm{MgSO}_{4}$ and concentrated to give the crude compound which was purified by flash chromatography ( $10 \% \mathrm{v} / \mathrm{v}$ ethyl acetate-hexane) to give the product as a colourless liquid $(1.69 \mathrm{~g}, 61 \%)$; $v_{\text {max }}($ neat $) / \mathrm{cm}^{-1} 3053,3000,2926,1628$, $1595,1580,1576,1508,1465,1440,1349,1271,1241,1179$, 1157, 1109, 1101, 1083, 1069, 1020, 999, 960, 916, 862, 841, 793, $772,728,635,614 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 2.83(1 \mathrm{H}, \mathrm{dd}, J 2.6$, 4.9, CH-O-CHH), 2.94 ( $1 \mathrm{H}, \mathrm{t}, J 4.2$, CH-O-CHH), 3.45-3.50 ( $\left.1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{O}-\mathrm{CH}_{2}-\mathrm{CH}-\mathrm{O}-\mathrm{CH}_{2}\right), 4.11(1 \mathrm{H}, \mathrm{dd}, J 5.6,11.1$, Ar-O-CHH-CH-O-CH2), 4.37 ( 1 H , dd, J 3.2, 11.1, Ar-$\mathrm{O}-\mathrm{CH} H-\mathrm{CH}-\mathrm{O}-\mathrm{CH}_{2}$ ), 6.77-6.79 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$ ), 7.32-7.52 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}$ ), 7.75-7.82 ( $1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 8.27-8.32(1 \mathrm{H}, \mathrm{m}$, $\mathrm{Ar}-\mathrm{H}) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 45.17\left(\mathrm{CH}_{2}\right), 50.67\left(\mathrm{CH}_{2}\right), 69.34$ $(C H), 105.36,121.26,122.44,125.75,125.98,126.14,126.94$, 127.88, 134.92, 154.63 (Ar-ipso); m/z (CI) 201 ( $[\mathrm{M}+\mathrm{H}]^{+}$, $100 \%$ ), 200 (13), 185 (53), 144 (17), 115 (10).

## ( $\pm$ )- $N$-Isopropyl- $N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine 24

A solution of the epoxide $(0.2 \mathrm{~g}, 1.0 \mathrm{mmol})$ in acetonitrile $(1 \mathrm{~mL})$ was treated with anhydrous metal salt $\left(\mathrm{CaCl}_{2}\right)(0.11 \mathrm{~g}$, 1.0 mmol ), then stirred until the salt had completely dissolved. The resulting solution was treated under stirring, at room temperature, with the required amount of the amine $(0.12 \mathrm{~g}, 2.0$ mmol ). After the addition of the amine was complete, the reaction mixture was stirred until the reagents had been consumed.

The reaction was then treated with water $(5 \mathrm{~mL})$ and extracted with diethyl ether ( $3 \times 10 \mathrm{~mL}$ ). The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was recrystallised with ether to give ( $\pm$ )- N -isopropyl-$N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine 24 as a white solid ( $0.24 \mathrm{~g}, 93 \%$ ). Mp 88-89 ${ }^{\circ} \mathrm{C}$; $v_{\max }\left(\mathrm{CDCl}_{3}\right) / \mathrm{cm}^{-1} 3429,2253$, $1655,1581,1461,1396,1268,1103,908,734,650 ; \delta_{\mathrm{H}}(300 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 1.11\left(6 \mathrm{H}, \mathrm{d}, J 6.4, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.51(2 \mathrm{H}, \mathrm{br} \mathrm{s}, \mathrm{OH}$, $\mathrm{N} H), 2.82-2.90\left(2 \mathrm{H}, \mathrm{m}, \mathrm{C} H_{2} \mathrm{NH}\right), 2.89-3.03(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right)$, 4.11-4.24 (3 H, m, CH2CH(OH)), $6.84(1 \mathrm{H}, \mathrm{dd}$, $J 1.2,7.4$, Ar-H), 7.34-7.52 (4 H, m, Ar-H), 7.76-7.83 ( $1 \mathrm{H}, \mathrm{m}$, $\mathrm{Ar}-\mathrm{H}), 8.21-8.26(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}) ; \delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 23.51$ $\left(\mathrm{CH}_{3}\right), 23.65\left(\mathrm{CH}_{3}\right), 49.31(\mathrm{CH}), 49.85\left(\mathrm{CH}_{2}\right), 68.98(\mathrm{CH})$, $71.06\left(\mathrm{CH}_{2}\right), 105.28,121.00,122.21,125.65,126.23,126.83$, 127.93 (Ar-ipso); m/z (CI) 260 ( $\mathrm{M}+\mathrm{H}]^{+}, ~ 77 \%$ ), 242 (7), 145 (34), 128 (6), 116 (74), 100 (100), 98 (28), 84 (13), 72 (31), 58 (30) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 260.1646 . \mathrm{C}_{16} \mathrm{H}_{21} \mathrm{NO}_{2}$ requires $\mathrm{m} / \mathrm{z}$, 260.1650).

## $N$-tert-Butoxycarbonyl- $N$-isopropyl- $N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine rac-22

To a solution of ( $\pm$ )- $N$-isopropyl- $N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine ( $0.95 \mathrm{~g}, 3.66 \mathrm{mmol}$ ) in dichloromethane ( 40 mL ), triethylamine ( $1.85 \mathrm{~g}, 18.30 \mathrm{mmol}$ ) followed by di-tert-butyl dicarbonate $(1.60 \mathrm{~g}, 7.32 \mathrm{mmol})$ were added. The reaction mixture was stirred overnight, then saturated ammonium chloride solution ( 30 mL ) was added. The phases were separated and the aqueous layer extracted with dichloromethane $(2 \times 40 \mathrm{~mL})$. The combined organic layers were dried over magnesium sulfate, filtered, and the solvent removed in vacuo. Purification by flash chromatography on silica with $10 \%$ ethyl acetate-hexane, gave the product 22 as a colourless liquid (1.45 $\mathrm{g}, 98.7 \%$ ); $v_{\text {max }}$ (neat) $/ \mathrm{cm}^{-1} 3416,3053,2975,2931,2876,1865$, $1658,1596,1581,1509,1476,1457,1403,1367,1347,1269$, 1241, 1213, 1164, 1128, 1103, 1069, 1020, 1001, 922, 900, 878, $861,792,771,735,571 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.15(3 \mathrm{H}, \mathrm{d}, J 6.6$, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{3}\right), 1.24\left(3 \mathrm{H}, \mathrm{d}, J 6.9, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{3}\right), 1.51(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 3.52\left(2 \mathrm{H}, \mathrm{d}, \mathrm{J} 4.3, \mathrm{CH}_{2}-\mathrm{N}(\mathrm{tBoc})\right), 4.05-4.07(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 4.17-4.25\left(3 \mathrm{H}, \mathrm{m}, \mathrm{C} \mathrm{H}_{2} \mathrm{CH}(\mathrm{OH})\right), 5.29(1 \mathrm{H}, \mathrm{br} \mathrm{s}$, $\mathrm{OH}), 6.85(1 \mathrm{H}, \mathrm{d}, J 7.4, \mathrm{Ar}-\mathrm{H}), 7.35-7.52(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 7.78-$ $7.85(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 8.19-8.24(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}) ; \delta_{\mathrm{C}}(75.5 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 20.89\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{3}\right), 21.40\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{3}\right),} 28.86\right.$ $\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 47.46(\mathrm{CH}), 66.24\left(\mathrm{CH}_{2}\right), 70.24\left(\mathrm{CH}_{2}\right), 72.50(\mathrm{CH})$, $81.17\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 105.16,120.99,122.04,125.64,125.82,126.30$, 126.80, 127.99, 134.89 (Ar-ipso), 154.53 (C=O); m/z (FAB) 360 ( $[\mathrm{M}+\mathrm{H}]^{+}, 12 \%$ ), 304 (20), 260 (42), 160 (100), 154 (25), 136 (20) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 360.2170 . \mathrm{C}_{21} \mathrm{H}_{29} \mathrm{NO}_{4}$ requires $\mathrm{m} / \mathrm{z}$, 360.2174).

## N -tert-Butoxycarbonyl- N -isopropyl- N -[2-oxo-3-(1-naphthyloxy)propyl] amine 23

$N$-tert-Butoxycarbonyl- $N$-isopropyl- $N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine $22(1.40 \mathrm{~g}, 3.89 \mathrm{mmol})$ was added to a solution of pyridinium dichromate ( $5.13 \mathrm{~g}, 13.63 \mathrm{mmol}$ ) in anhydrous $N, N$-dimethylformamide ( 30 mL ). The reaction mixture was stirred overnight, then worked up with brine ( 40 $\mathrm{mL})$. The compound was extracted with diethyl ether ( $3 \times 40$ mL ). The combined organic layers were dried over anhydrous magnesium sulfate, filtered and concentrated under reduced pressure to give the crude compound which was purified by flash chromatography eluting with $10 \%$ ethyl acetate-hexane to give N -tert-butoxycarbonyl- N -isopropyl- N -[2-oxo-3-(1-naphthyloxy)propyl]amine 23 ( $0.95 \mathrm{~g}, 68.5 \%$ ); $v_{\max }$ (neat)/ $\mathrm{cm}^{-1} 3054$, 2976, 2930, 1742, 1699, 1695, 1651, 1597, 1581, 1509, 1464, 1398, 1365, 1270, 1243, 1214, 1165, 1105, 1077, 1021, 902, 875, 792, 771; $\delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.06,1.09(6 \mathrm{H}, 2 \times \mathrm{dd}$, ratio $\left.60: 40, J 6.8,6.9, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.41,1.49(9 \mathrm{H}, 2 \times \mathrm{s}$, ratio $\left.60: 40, \mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 4.21,4.25\left(2 \mathrm{H}, 2 \times \mathrm{s}\right.$, ratio $\left.60: 40, \mathrm{CH}_{2}\right)$,
4.44-4.65 (1 H, m, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 4.80,4.88(2 \mathrm{H}, 2 \times \mathrm{s}$, ratio $\left.60: 40, \mathrm{CH}_{2}\right), 6.71-6.75(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 7.34-7.56(4 \mathrm{H}, \mathrm{m}$, Ar-H), 7.79-7.87 (1 H, m, Ar-H), 8.30-8.33 (1 H, m, Ar-H); $\delta_{\mathrm{C}}\left(75.5 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 20.88,21.30\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 28.69,28.84$ $\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 46.44(\mathrm{CH}), 49.13\left(\mathrm{CH}_{2}\right), 72.67\left(\mathrm{CH}_{2}\right), 80.54$ $\left(C\left(\mathrm{CH}_{3}\right)_{3}\right), 105.25,121.80,122.10,126.16,127.06,128.08$, 135.03 (Ar-ipso); m/z (FAB) 357 ([M] ${ }^{+}, 64 \%$ ), 339 (16), 313 (18), 302 (98), 289 (14), 258 (61), 158 (59), 154 (100), 136 (95), 116 (58), 107 (65), 95 (79) (Found: $[M]^{+}, 357 . \mathrm{C}_{21} \mathrm{H}_{27} \mathrm{NO}_{4}$ requires $m / z, 357)$.

## (R)-(-)-N-tert-Butoxycarbonyl- $N$-isopropyl- $N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine 22

A mixture of ( $p$-cymene)ruthenium(II) chloride dimer $(0.17 \mathrm{mg}$, $0.0028 \mathrm{mmol})$ and $(1 R, 2 R)$-TsDPEN $2(0.21 \mathrm{mg}, 0.0021 \mathrm{mmol})$ in a $5: 2$ formic acid-triethylamine mixture $(2.5 \mathrm{~mL})$ was stirred at $28^{\circ} \mathrm{C}$ for 15 min . $N$-Isopropyl- $N$-[2-oxo-3-(1-naphthyloxy)propyl]amine $(0.37 \mathrm{~g}, 1.12 \mathrm{mmol})$ was added the solution was stirred at $28^{\circ} \mathrm{C}$ for 24 hours. Then the mixture was filtered through silica and washed with ethyl acetate ( 60 ml ). The solvent was evaporated under reduced pressure to give the crude compound, which was purified by flash chromatography ( $20 \%$ $\mathrm{v} / \mathrm{v}$ ethyl acetate-petroleum ether $40: 60)$ to give $R-(-)-22$ as a colourless liquid $(0.36 \mathrm{~g}, 98.4 \%) ;[\alpha]_{\mathrm{D}}^{20}-2.92(c=0.96$, ethanol); $v_{\max }($ neat $) / \mathrm{cm}^{-1} 3418,3053,2974,2931,1738,1687,1596,1581$, $1509,1403,1366,1269,1164,1103,1069,1020,1001,899$, $861,792,771,736 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.15(3 \mathrm{H}, \mathrm{d}, J 6.6$, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{3}\right), 1.24\left(3 \mathrm{H}, \mathrm{d}, J 6.8, \mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{3}\right), 1.51(9 \mathrm{H}, \mathrm{s}$, $\left.\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 3.52\left(2 \mathrm{H}, \mathrm{d}, J 4.5, \mathrm{CH}_{2}-\mathrm{N}(\mathrm{tBoc})\right), 4.04-4.11(1 \mathrm{H}, \mathrm{m}$, $\left.\mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 4.17-4.25\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}(\mathrm{OH})\right), 5.17(1 \mathrm{H}, \mathrm{br} \mathrm{s}$, $\mathrm{OH}), 6.85(1 \mathrm{H}, \mathrm{d}, J 7.4$, Ar-H), $7.34-7.52(4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 7.77-$ $7.84(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 8.19-8.25(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}) ; \delta_{\mathrm{C}}(75.5 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 20.88\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{3}\right), 21.39\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{3}\right), 28.86}\right.$ $\left(\mathrm{C}\left(\mathrm{CH}_{3}\right)_{3}\right), 49.08(\mathrm{CH}), 66.27\left(\mathrm{CH}_{2}\right), 70.21\left(\mathrm{CH}_{2}\right), 72.90(\mathrm{CH})$,
 126.80, 127.99, 134.89 (Ar-ipso), 154.53 (C=O); m/z (CI) 360 $\left([\mathrm{M}+\mathrm{H}]^{+}, 13 \%\right), 304$ (21), 260 (36), 160 (98), 136 (19), 133 (100) (Found: $[\mathrm{M}+\mathrm{H}]^{+}$, 360.2177. $\mathrm{C}_{21} \mathrm{H}_{29} \mathrm{NO}_{4}$ requires $\mathrm{m} / \mathrm{z}$, 360.2175 ). The reduction using ( $1 S, 2 R$ )-1-propan-2-ol was carried out using identical conditions to those reported. ${ }^{4 b}$ The product, $R-(-)-\mathbf{2 2}$ was formed in $99 \%$ yield and $64 \%$ ee as determined by chiral HPLC analysis.

## (R)-(+)-N-Isopropyl- $N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine 24

To a solution of $(R)$ - $N$-tert-butoxycarbonyl- $N$-isopropyl- $N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine $R-(-)-22(0.32 \mathrm{~g}, 0.90$ $\mathrm{mmol})$ in dichloromethane ( 2 mL ) TFA ( 2 mL ) was added slowly. After 2 hours the solvent was removed and the resulting product was dissolved in ethyl acetate and NaOH solution 0.2 M was added until the $\mathrm{pH}=7$. The product was then extracted with ethyl acetate and the combined organic layers were dried over magnesium sulfate, filtered and the solvent removed in vacuo. Recrystallisation (ethyl acetate-petroleum ether $40: 60$ ) gave $(R)-N$-isopropyl- $N$-[2-hydroxy-3-(1-naphthyloxy)propyl]amine $R-(+)$ - 24 as a light yellow solid $(0.20 \mathrm{~g}, 84.5 \%$ yield). The product was determined to be of $83.0 \%$ ee by HPLC analysis (Chiral OD, hexane-ethanol-diethylamine = 95:5:0.1( $\left.0.5 \mathrm{~mL} \mathrm{~min}^{-1}\right), R$ isomer $32.91 \mathrm{~min}, S$ isomer 54.76 $\min ), \operatorname{mp} 88-89^{\circ} \mathrm{C} ;[a]_{\mathrm{D}}^{20}+5.1(c=1.6$, ethanol $) ; v_{\text {max }}\left(\mathrm{CDCl}_{3}\right) /$ $\mathrm{cm}^{-1} 3429,2253,1655,1581,1461,1396,1268,1103,908,734$, $650 ; \delta_{\mathrm{H}}\left(300 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) 1.11\left(6 \mathrm{H}, \mathrm{d}, J 6.4, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 2.14$ ( 2 H , br s, $\mathrm{OH}, \mathrm{NH}$ ), 2.81-2.91 ( $2 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{NH}$ ), 2.98-3.04 ( $\left.1 \mathrm{H}, \mathrm{m}, \mathrm{CH}\left(\mathrm{CH}_{3}\right)_{2}\right), 4.11-4.21\left(3 \mathrm{H}, \mathrm{m}, \mathrm{CH}_{2} \mathrm{CH}(\mathrm{OH})\right), 6.83$ ( 1 H , dd, $J 0.9,7.5$, Ar-H), 7.34-7.53 ( $4 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 7.78-$ $7.83(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}), 8.21-8.28(1 \mathrm{H}, \mathrm{m}, \mathrm{Ar}-\mathrm{H}) ; \delta_{\mathrm{C}}(75.5 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) 23.45\left(\mathrm{CH}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{3}\right)$, $23.59\left(\mathrm{C}\left(\mathrm{CH}_{3}\right) \mathrm{CH}_{3}\right), 49.36$ $(\mathrm{CH}), 49.83\left(\mathrm{CH}_{2}\right), 69.81(\mathrm{CH}), 71.04\left(\mathrm{CH}_{2}\right), 80.05\left(\mathrm{C}_{\left.\left(\mathrm{CH}_{3}\right)_{3}\right) \text {, }}\right.$ 105.27, 121.01, 122.21, 125.65, 126.23, 126.84, 127.93 (Ar-ipso);
$m / z(\mathrm{CI}) 260\left([\mathrm{M}+\mathrm{H}]^{+}, 77 \%\right), 242$ (7), 145 (34), 128 (6), 116 (74), 100 (100), 98 (28), 84 (13), 72 (31), 58 (30) (Found: $[\mathrm{M}+\mathrm{H}]^{+}, 260.1650 . \mathrm{C}_{16} \mathrm{H}_{21} \mathrm{NO}_{2}$ requires $m / z, 260.1650$ ). The compound from reduction of 23 using $\mathrm{Ru}(\mathrm{II})$-aminoindanol 1 was deprotected in an identical manner to give $R-(+)$-24 in $93 \%$ yield and $64 \%$ ee.

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[^0]:    $\dagger$ On leave from the Brazilian Space Aeronautical Institute-Praca Mal. Eduardo Gomes 50, Vila das Acacias, SJ Campos, SP, Brazil.

